

## Supplementary Material

**Effect of Substituents in directing the formation of Benzochlorins and isobacteriochlorins in Porphyrin and chlorin systems:**

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Grossman Thomas Dougherty and Ravindra K Pandey\*

## CRYSTAL DATA AND STRUCTURE REFINEMENT

Compound name	2 <sup>1</sup> -trifluoromethyl-octaethy benzochlorin
Color/shape	deep red/ rectangular prism
Empirical Formula	C <sub>40</sub> H <sub>47</sub> N <sub>4</sub> F <sub>4</sub>
Formula weight	640
Temperature	22±3°C
Crystal system	Triclinic
Space group	P 1
Unit cell dimensions	a (Å) 8.689 (3) α= 73.51 (2)° b (Å) 14.490 (4) β=81.50(3)° c (Å) 14.953 (5) γ= 77.80(2)°
Volume	1752.4 (9) (Å <sup>3</sup> )
Z	2 ( two per unit cell )
Density calculated	1.213 mg/m <sup>3</sup>
Absorption coefficient	0.81 mm <sup>-1</sup>
Diffractionmeter	CAD-4
Radiation/wavelength	CuKα / 1.5418Å
F(000)	684
Crystal size	0.25x0.35x0.15 mm
θ range for data collection	0 to 150°
Index ranges	-10 ≤ h ≤ 10; 0 ≤ k ≤ 17; 18 ≤ l ≤ 18
Reflections collected	7109
Independent/observed reflections	2569 ( I ≤ 3σ )
Absorption correction	semi-empirical from psi scans
Range of relat. trasm. factors	0.99 and 0.90
Refinement method	Full-matrix least squares on F <sup>2</sup>
Computing	SHELXS-96
Data / restraints/ parameters	2569 /0/1248
Goodness of fit on F <sup>2</sup>	1.017
Function minimized	Σ [   F <sub>o</sub> <sup>2</sup>   - (1/k)   F <sub>c</sub> <sup>2</sup>   ]
Final R indices [ I > 3σ (I) ]	0.0825
Final extinction coefficient	4.41 x 10 <sup>-7</sup>
Large diff. peaks and hole	± 0.15 e/Å <sup>3</sup>

**X-ray Crystallography.** Crystals of

2<sup>1</sup>-trifluoromethyloctaethylbenzochlorin (C<sub>40</sub>H<sub>47</sub>N<sub>4</sub>F<sub>6</sub>) were obtained by a slow evaporation of the compound dissolved in methylene chloride and acetone mixture. The crystals are dark red in color and have a rectangular prismatic habit. A crystal of dimensions 0.25 × 0.35 × 0.15 mm was used for the present crystal structural investigation. The crystals belong to the triclinic system, space group *P*1 (two independent molecules) with the following cell dimensions:  $a = 8.689(3)$ ,  $b = 14.490(4)$ ,  $c = 14.953(5)$  Å,  $\alpha = 73.51(2)$ ,  $\beta = 81.50(3)$ ,  $\gamma = 77.80(2)^\circ$ ,  $V = 1752.4$  Å<sup>3</sup>,  $\rho_{\text{calc}} = 1.213$  g/cm<sup>3</sup>,  $\mu$  (Cu K $\alpha$ ) = 0.81 mm<sup>-1</sup> and  $Z = 2$ . Complete three-dimensional data was collected on a CAD-4 computer-controlled diffractometer. A total of 7109 reflections ( $2\theta_{\text{max}} = 150^\circ$ ) were collected by the  $\omega/2\theta$  method, out of which 2596 were considered significant ( $I \geq 3\sigma$ ). The crystal structure was solved by a routine and straightforward application of SHJELX-96 program on a silicon graphics R10,000 computer. The structure was refined using SHELX-96 package of programs. All the hydrogen atoms were located on a difference Fourier map. Refinements were carried out with anisotropic thermal parameters for the non-hydrogen atoms and isotropic thermal parameter for the hydrogen atoms. The final reliability index (*R* factor) was 0.0825 for the observed 2569 reflections, the goodness of fit parameter  $S = 1.267$ .

## Supporting Material

1. Figure A:  $^1\text{H}$  NMR spectrum of isobacteriochlorin **9**
2. Figure Aa: Mass spectrum of isobacteriochlorin **9**
3. Figure B:  $^1\text{H}$  NMR spectrum of isobacteriochlorin **10**
4. Figure Bb: Mass spectrum of isobacteriochlorin **10**
5. Figure C:  $^1\text{H}$  NMR spectrum of isobacteriochlorin **11**
6. Figure Cc: Mass spectrum of isobacteriochlorin **11**.
7. Figure D:  $^1\text{H}$  NMR spectrum of porphyrin **12**
8. Figure E: Mass spectrum of porphyrin **12**
9. Figure F:  $^1\text{H}$  NMR spectrum of Benzochlorin **13**.
10. Figure G: Mass spectrum of Benzochlorin **13**
11. Figure H:  $^{19}\text{F}$  MR spectrum of Benzochlorin **13**
12. Figure I:  $^1\text{H}$  NMR spectrum of porphyrin **14**
13. Figure J: Mass spectrum of porphyrin **14**
14. Figure K:  $^{19}\text{F}$  MR spectrum of porphyrin **14**
15. Figure L:  $^1\text{H}$  NMR spectrum of chlorin **15a**
16. Figure M: Mass spectrum of chlorin **15a**
17. Figure N:  $^{19}\text{F}$  MR spectrum of chlorin **15a**
18. Figure O:  $^1\text{H}$  NMR spectrum of isobacteriochlorin **17**
19. Figure P: Mass spectrum of isobacteriochlorin **17**.
20. Figure Q:  $^{19}\text{F}$  MR spectrum of isobacteriochlorin **17**
21. Figure R: COSY data for isobacteriochlorin **17**
22. Figure S: ROESY data for isobacteriochlorin **17**
23. Figure T:  $^1\text{H}$  NMR spectrum of chlorin **18**.
24. Figure U: Mass spectrum of chlorin **18**
25. Figure V:  $^{19}\text{F}$  MR spectrum of chlorin **18**
26. Figure W: COSY data for chlorin **18**
27. Figure X: ROESY data for chlorin **18**

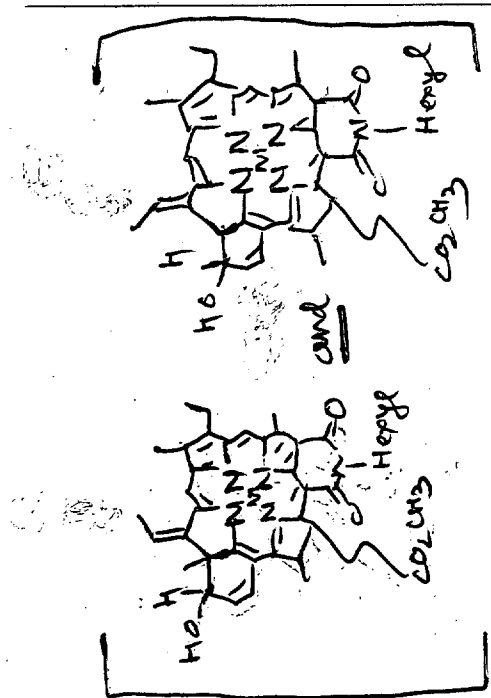
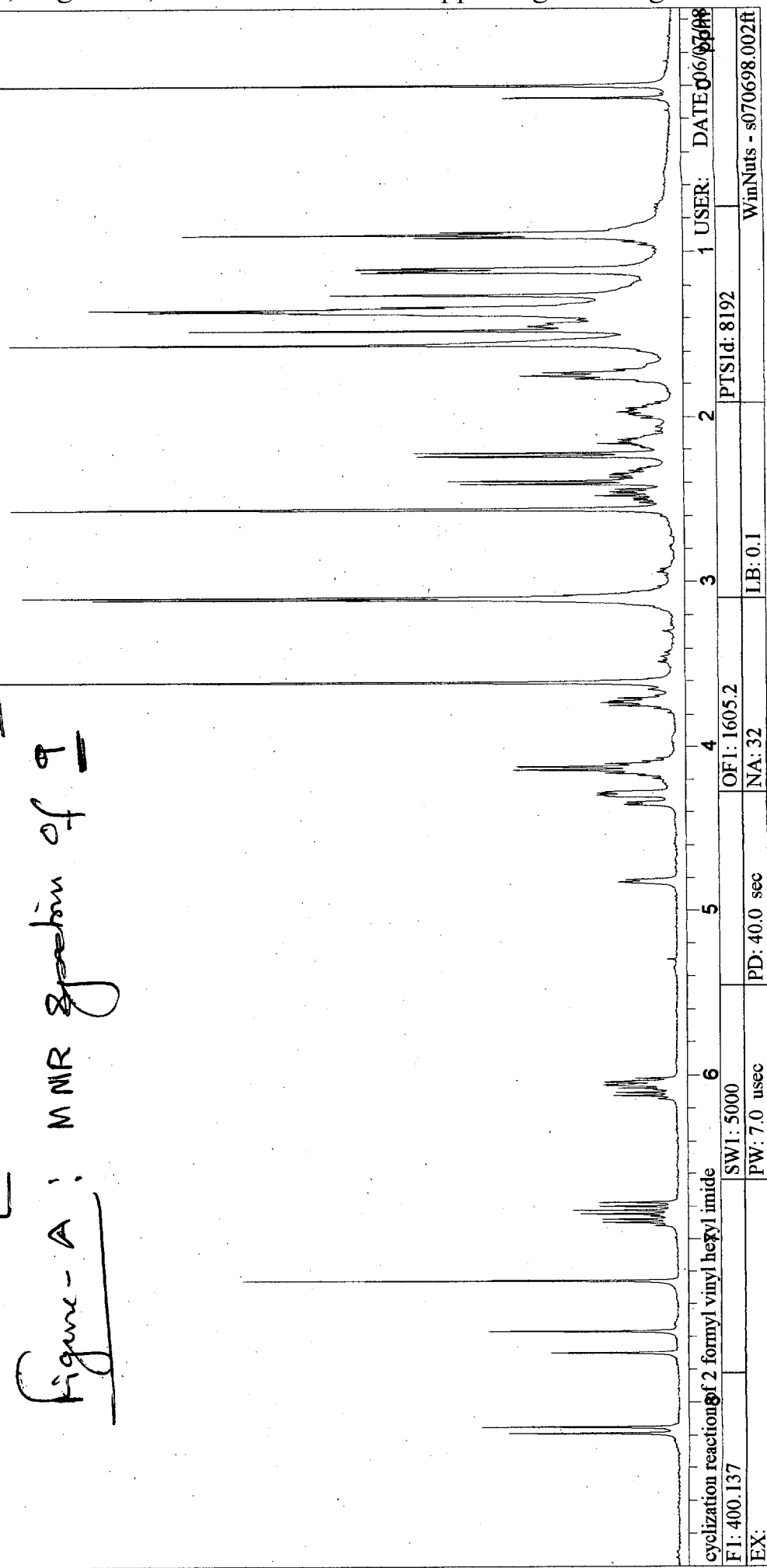


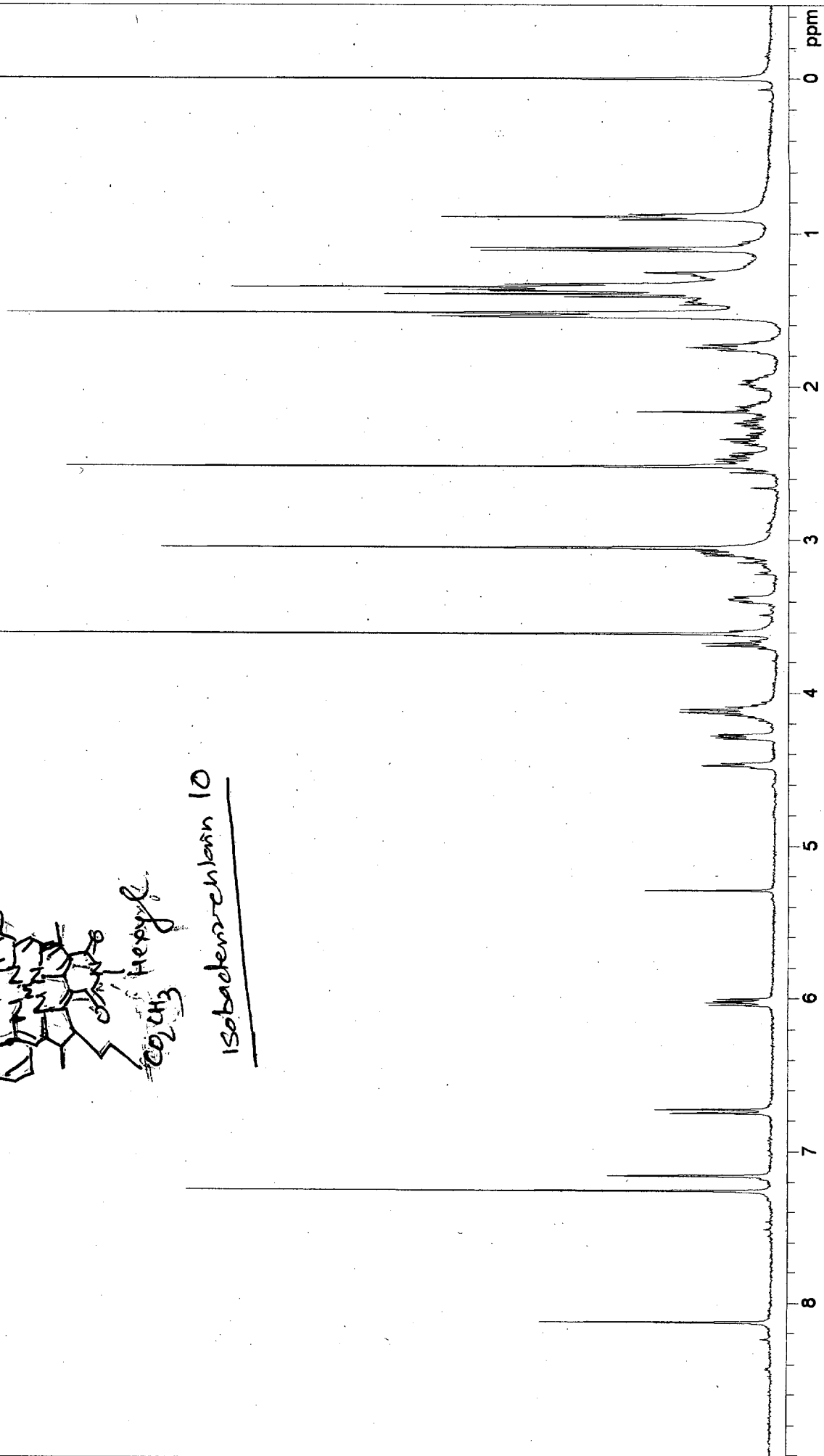
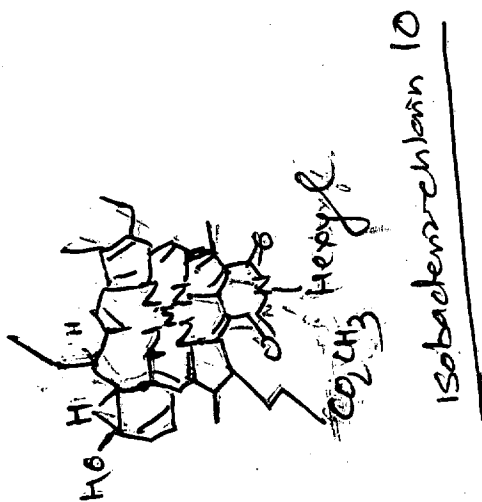
Figure - A : MNR spectrum of I



cyclization reaction of 2 formyl vinyl hexyl imide	SW1: 5000	OF1: 1605.2	1 USER: DATE: 06/07/08
FI: 400.137	PW: 7.0 usec	NA: 32	PTSId: 8192
EX:	PD: 40.0 sec	LB: 0.1	WinNuts - s070698.002ft

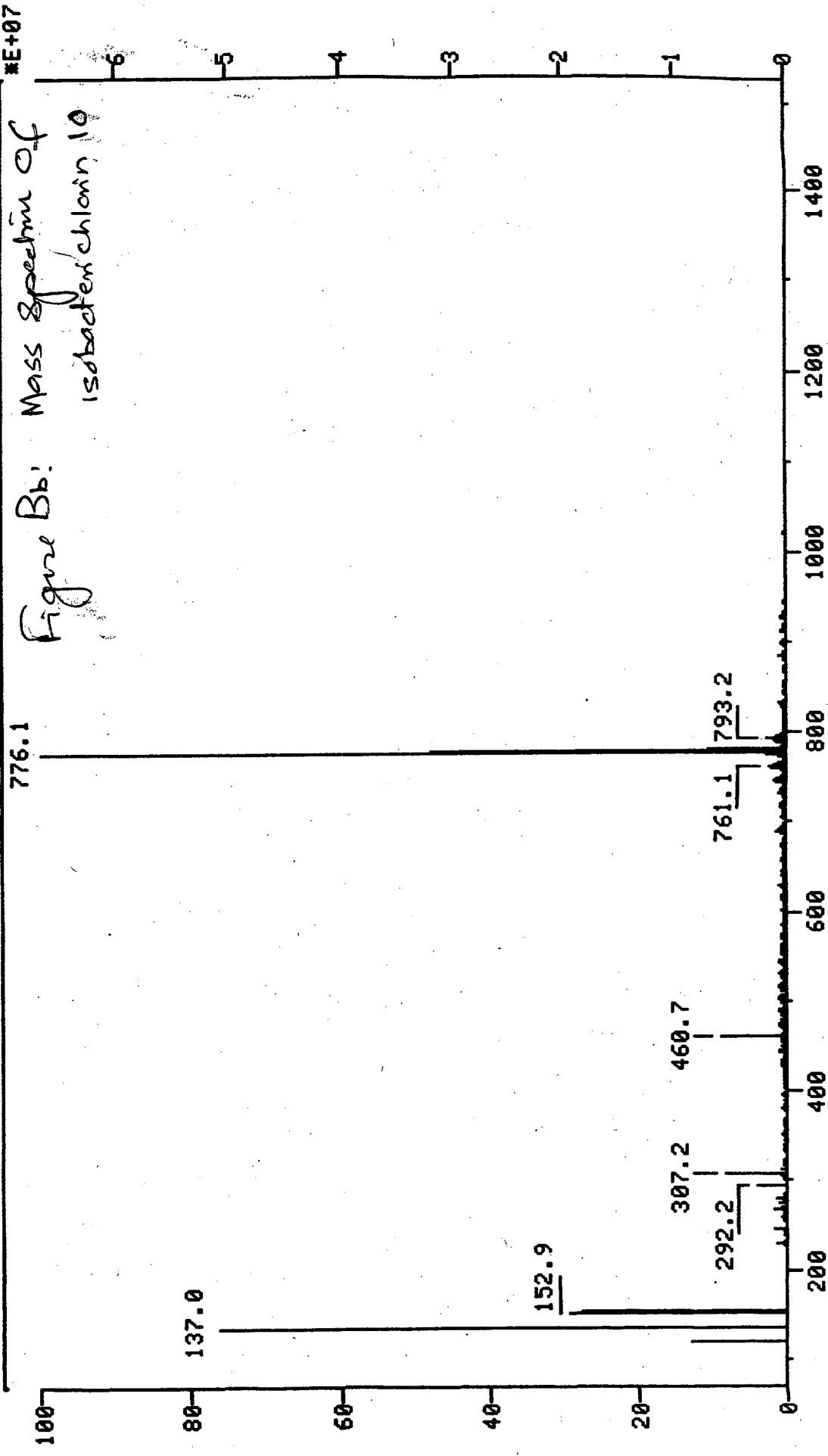
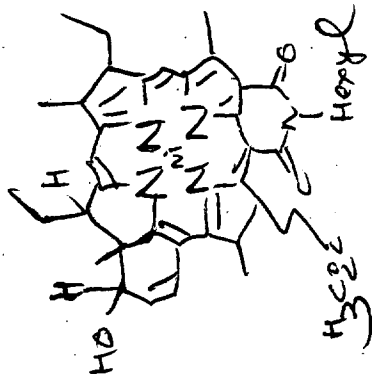


Figure B: <sup>1</sup>H NMR Spectrum of 10



P-202 monohydrogenated product of P-201	OF1: 2205.6	USER: DATE: 24/09/98
FI: 400.138	NA: 16	PTS1d: 8192
EX:	PD: 40.0 sec	WinNuts - s092498.003ft
	LB: 0.1	

SPEC: P202  
 Samp: P-202 (PANDEY) 6-OCT-98 DERIVED SPECTRUM 9  
 Comm: POS. FAB#m-NITROBENZYL ALCOHOL Start: 14:04:58 10  
 Mode: EI +VE +HMR BSCAN (EXP) UP LR  
 Oper: DUTTA  
 Base: 776.1 Inlet: 66777636 Masses: 100 > 1500  
 Norm: 776.1 RIC: 570630208 # peaks: 1259  
 Peak: 1000.00 mmu  
 Data: SY:P202.DT + 'P202' /1>10 - 'BLN10698



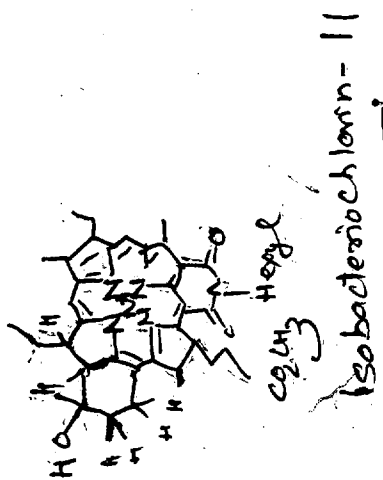
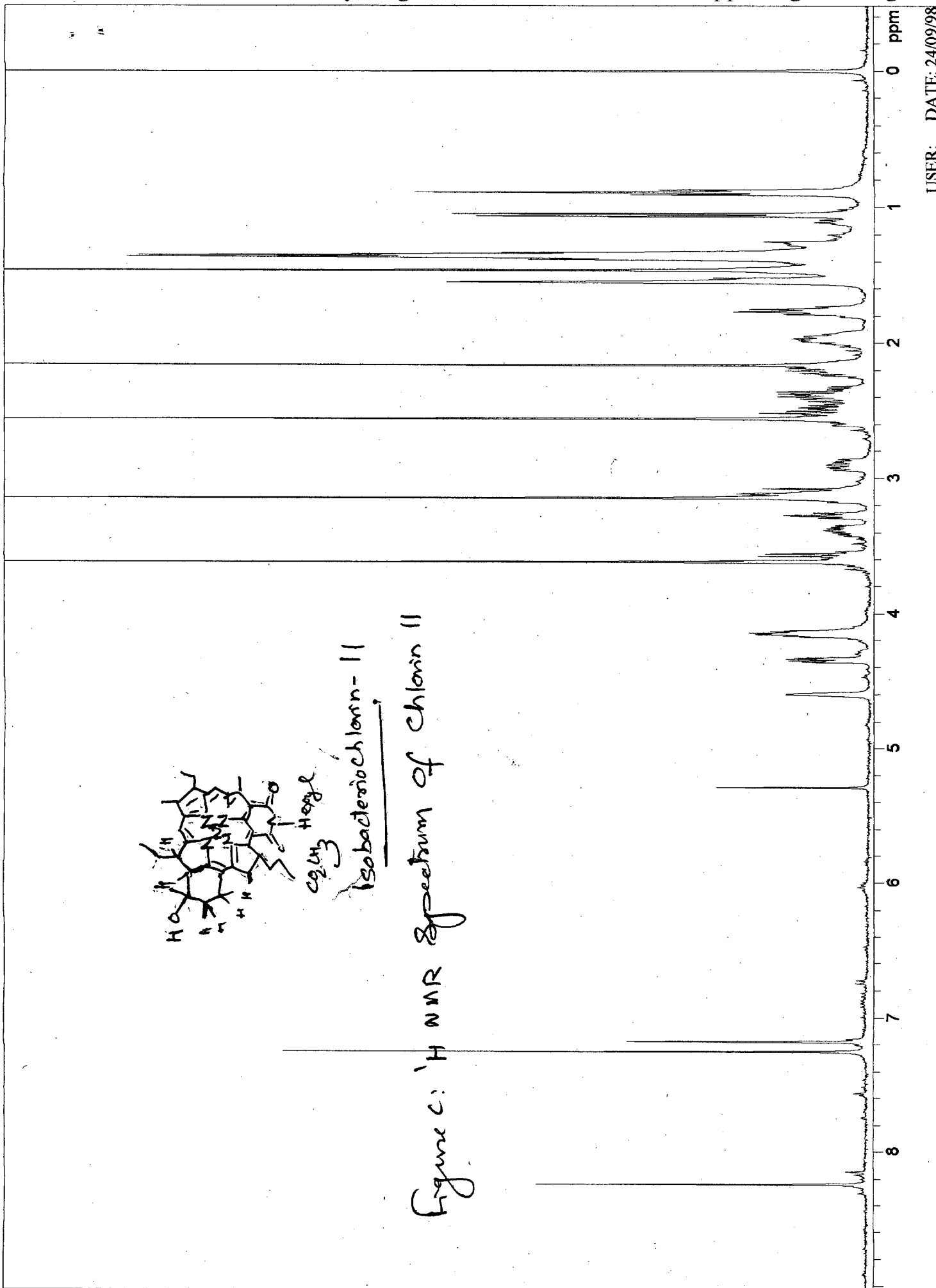


Figure c: <sup>1</sup>H NMR spectrum of chlorin II



USER: DATE: 24/09/98

FI: 400.138	SW1: 6024	OF1: 2205.5	PTSId: 8192
EX:	PW: 6.0 usec	NA: 16	WinNuts - s092498.004ft
	PD: 40.0 sec	LB: 0.1	



**SPEC:** P202 P-203  
**Samp:** P-202 (PANDEY)  
**Comm:** POS. FAB#m-NITROBENZYL ALCOHOL  
**Mode:** EI +VE +HMR BSCAN (EXP) UP LR.  
**Oper:** DUTTA  
**Base:** 778.2 Inten : 123745296  
**Norm:** 778.2 RIC : 944659968  
**Peak:** 1000.00 mmu  
**Data:** SY:P202.DT  
**Inlet :**  
**Masses:** 100 > 1200  
**# peaks:** 995  
**+ 'P202' /1>10 - 'BLN92498'**

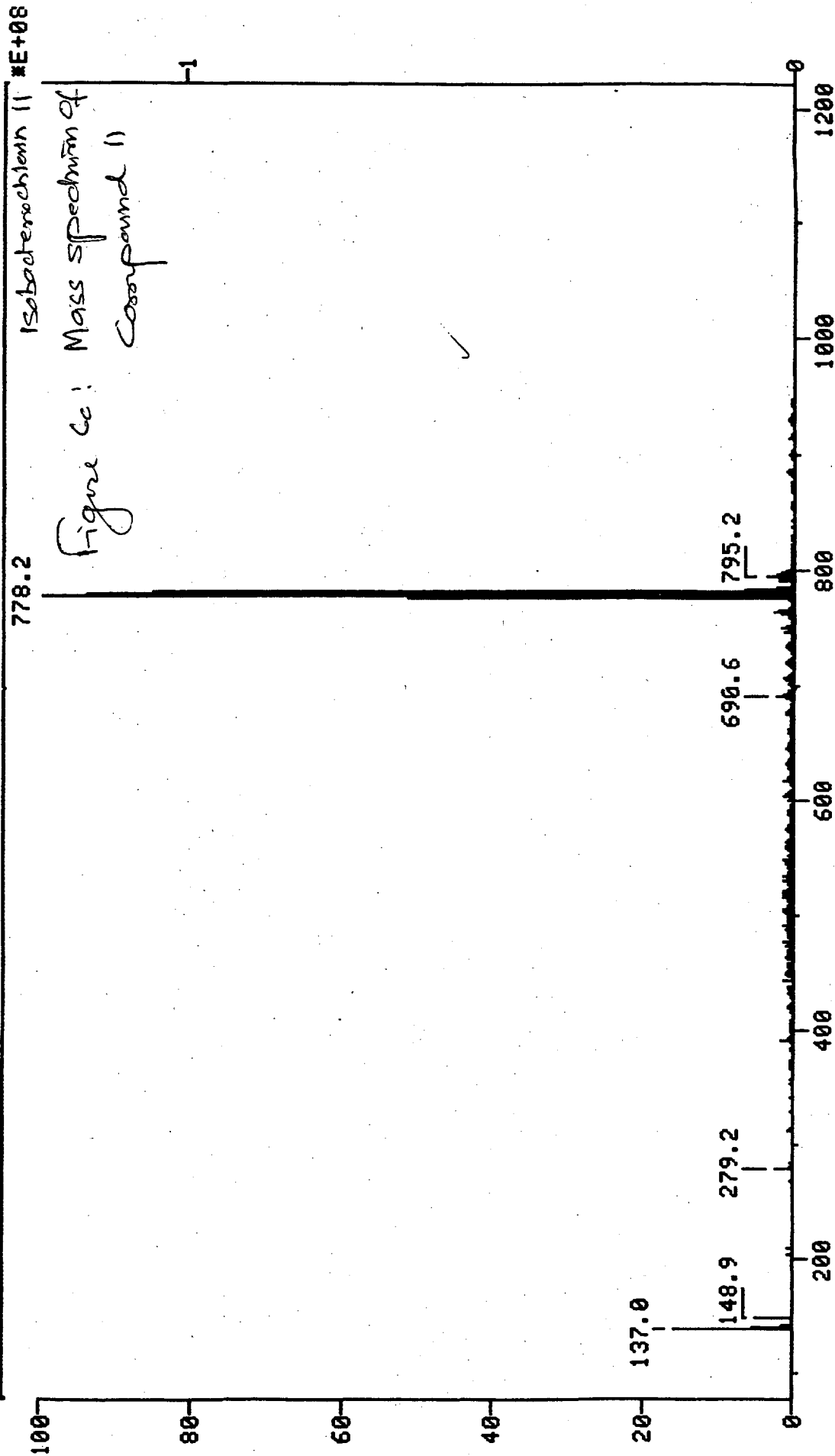
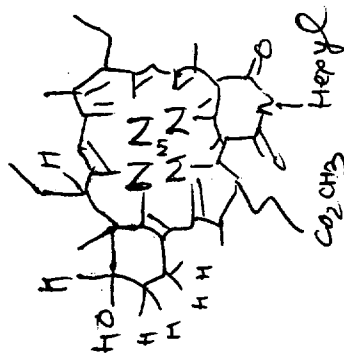
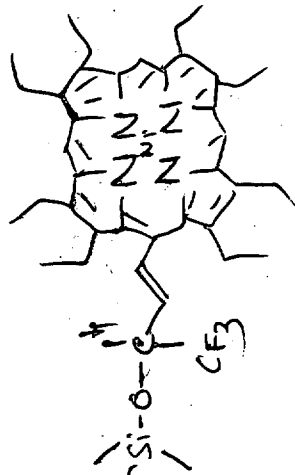


Figure-D



Porphyrim-12

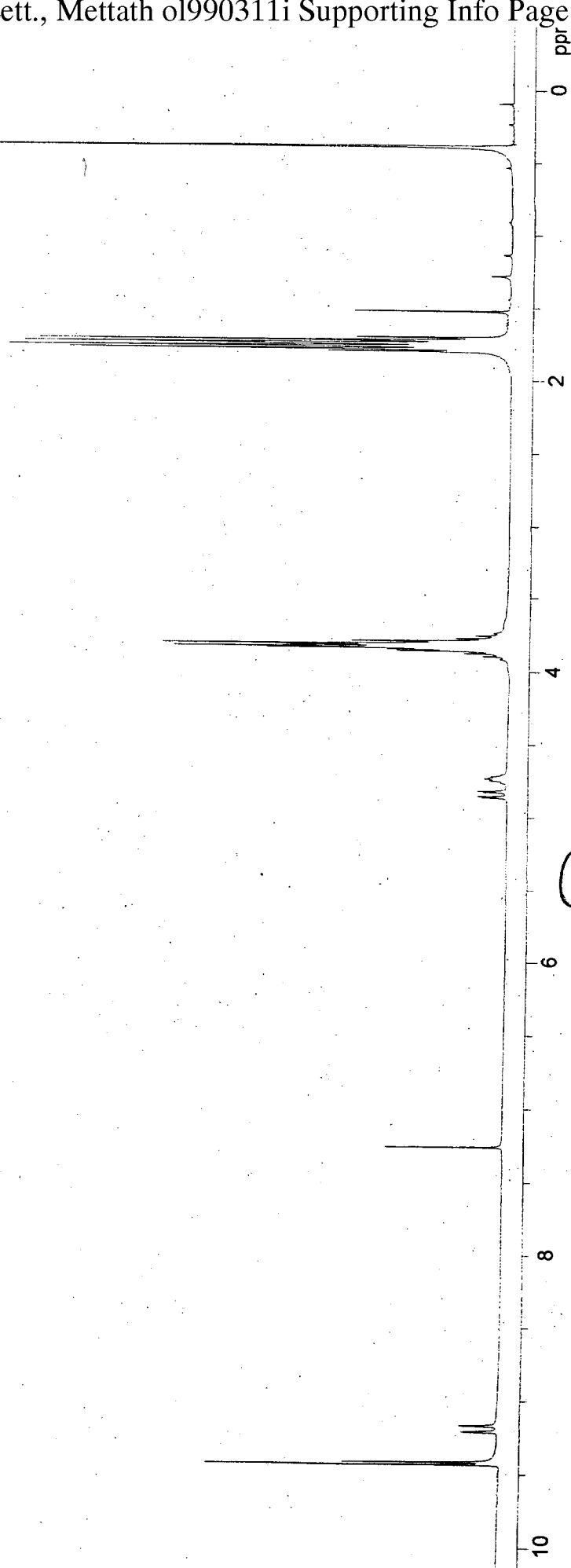


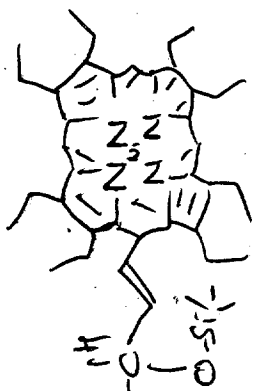
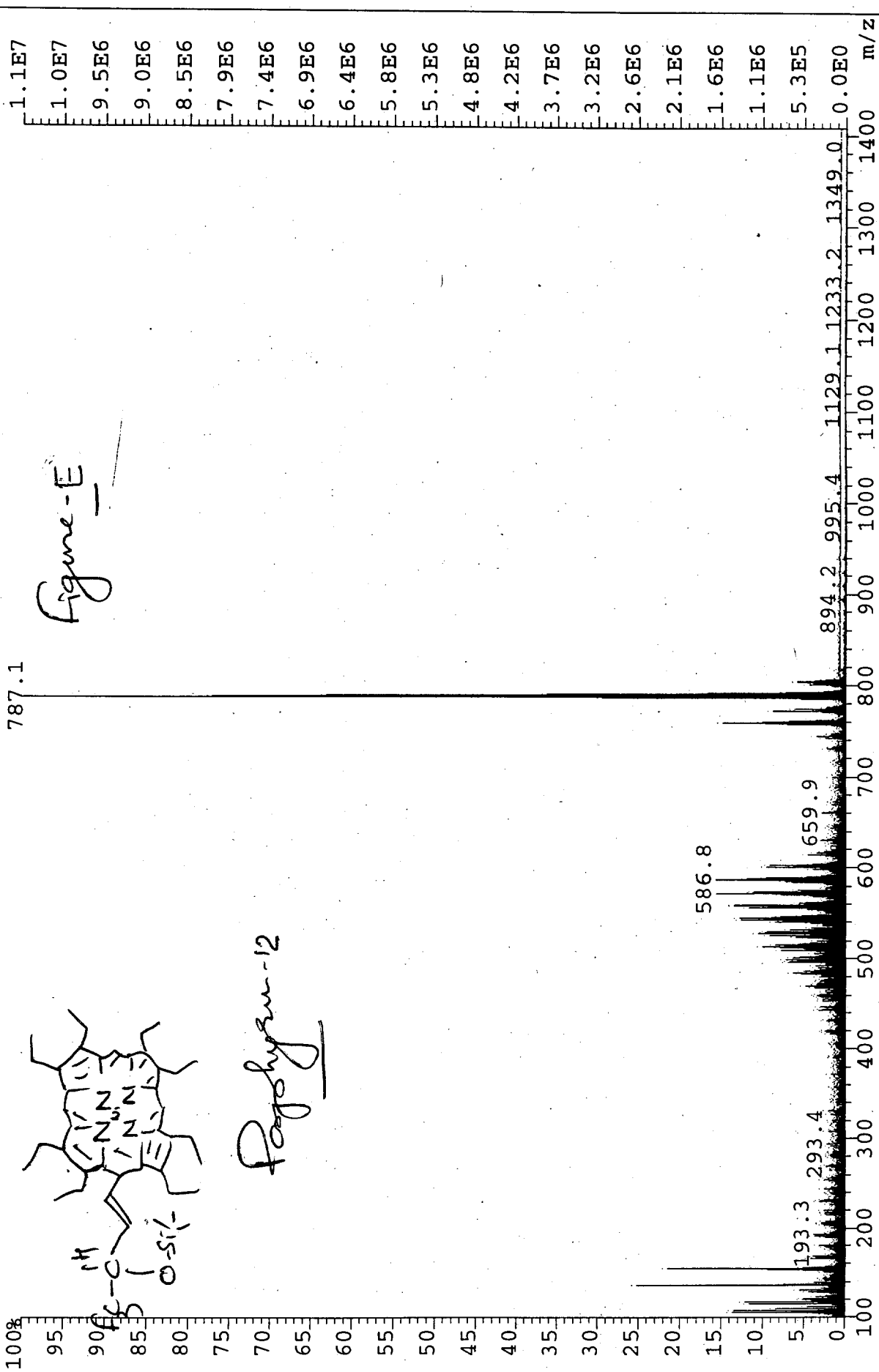
Figure-D: NMR Spectrum of 12

CDC13, FP-12  
 F1: 400.138  
 EX:

SW1: 6024  
 PW: 7.0 usec  
 OF1: 2206.2  
 NA: 32  
 PTSId: 8192  
 LB: 0.1  
 USER: DATE: 27/02/99  
 WinNuts - \$Fp12.001

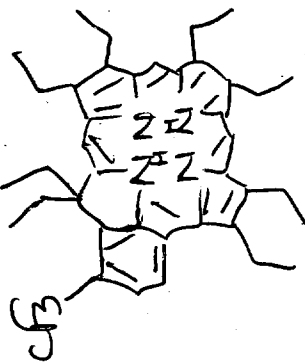
E

File:RP-FP-12A Ident:9\_12 Win 100PPM Acq:22-MAR-1999 17:30:47 +0:46 Cal:CF032299A\_1  
70S FAB+ Magnet BpM:787 BpI:10586112 TIC:290600160 Flags:NORM  
File Text:FP-12A



Porphyrin-12

Figure E. Mass Spectra of Porphyrin 12



Benzochlorin B

Figure - F

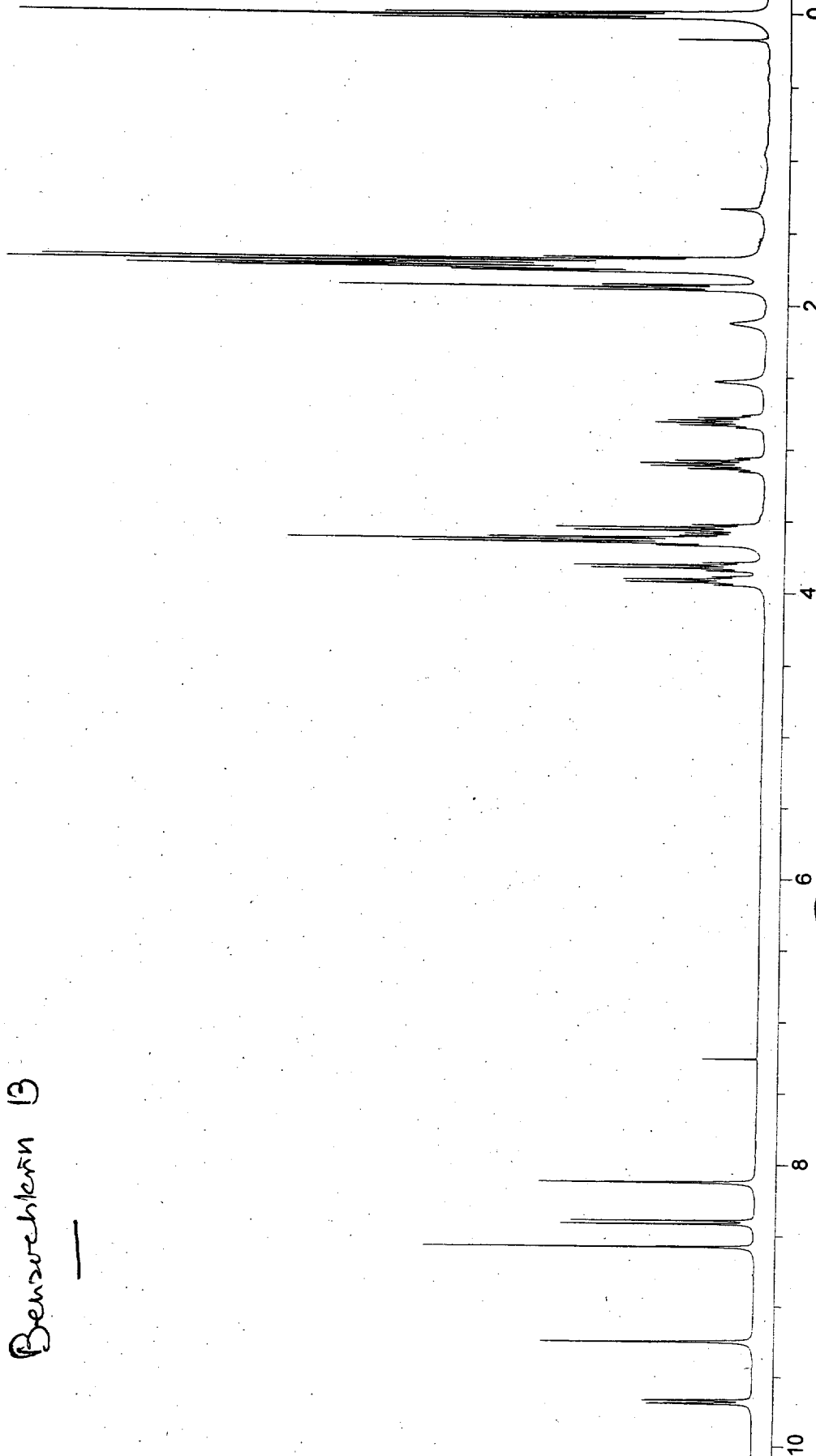
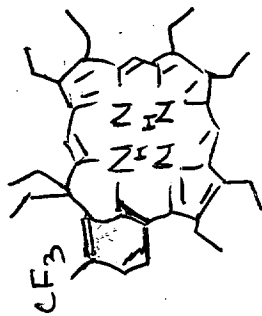


Figure-F- NMR Spectrum of Benzochlorin B

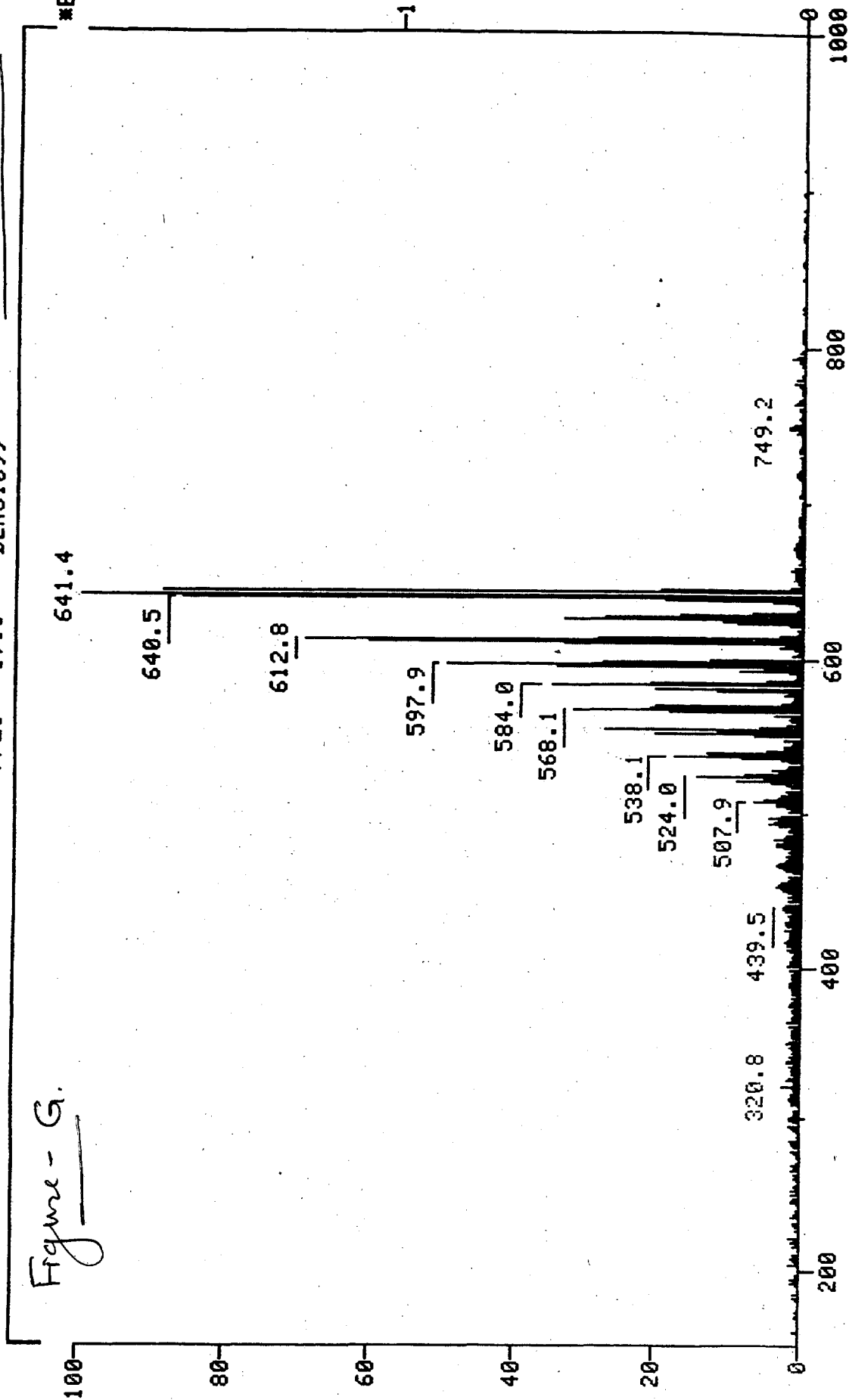
CDCB, FP-20	SWI: 8065	OF1: 2409.2	LB: 0.1	WinNuts - \$Fp20.002
FI: 400.138	PW: 10.0 usec	PD: 40.0 sec	PTSId: 8192	DATE: 02/04/99
EX:		NA: 32	13	

SPEC: FP20  
 Samp: FP-20 (PANDEY)  
 Comm: POS. FAB#m-NITROBENZYL ALCOHOL(1% TFA)  
 Mode: EI +VE +HMR BSCAN (EXP) UP LR  
 Oper: DUTTA  
 Base: 641.4  
 Norm: 641.4  
 Peak: 1000.00 mmu  
 Data: SY:FP20.DT

10-MAR-99 DERIVED SPECTRUM 9  
 Start : 10:34:22 10  
 Inlet :  
 Inten : 177974688 S  
 Masses: 150 > 2000  
 # peaks: 1981  
 + 'FP20' /1>10 - 'BLN31099'

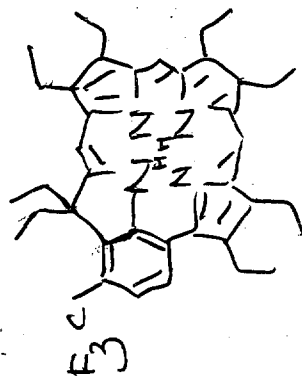


Benzochlorin - 13



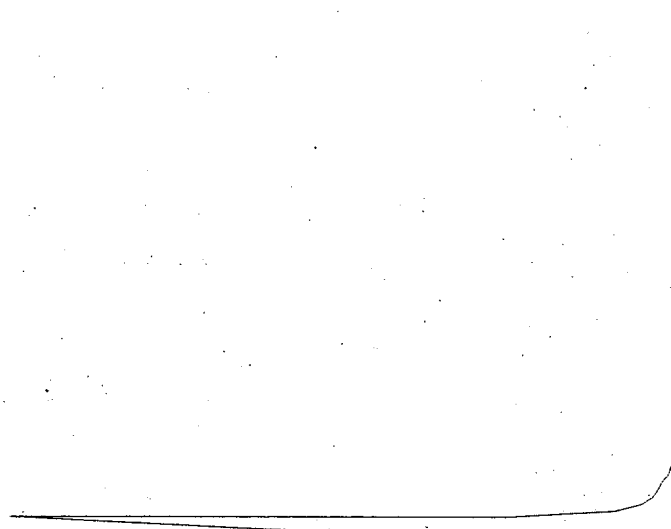
#E 8

<sup>19</sup>F NMR spectrum of Benzvalene - 13



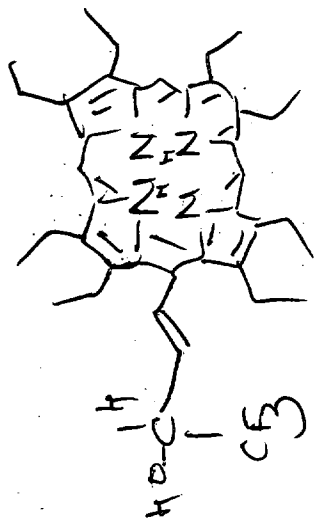
13

Figure - H

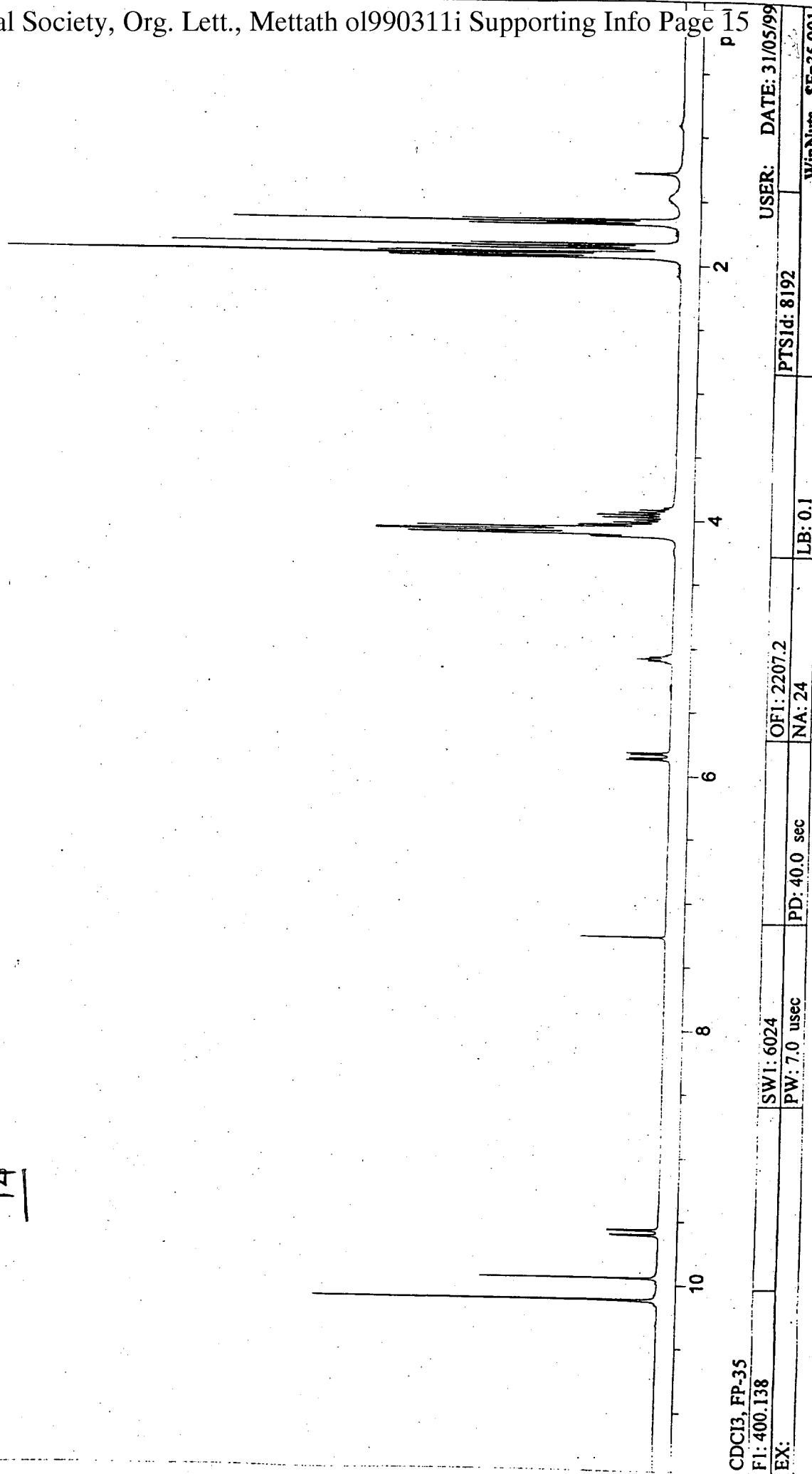


CDCl <sub>3</sub> , FP-20, F-19 NMR	SWH: 26315	OF1: -24724.4	IR: 2.0	USER: DATE: 03/03/99
EX: FI: 376.471	PW: 8.0 usec	PD: 20.0 sec	NA: 24	PTSID: 8192

Figure - I : <sup>1</sup>H NMR Spectrum of porphyrin 14.



14



CDCI3, FP-35

F1: 400.138

EX:

SW1: 6024

PW: 7.0 usec

PD: 40.0 sec

OF1: 2207.2

NA: 24

LB: 0.1

PTSId: 8192

USER:

DATE: 31/05/99

P 15

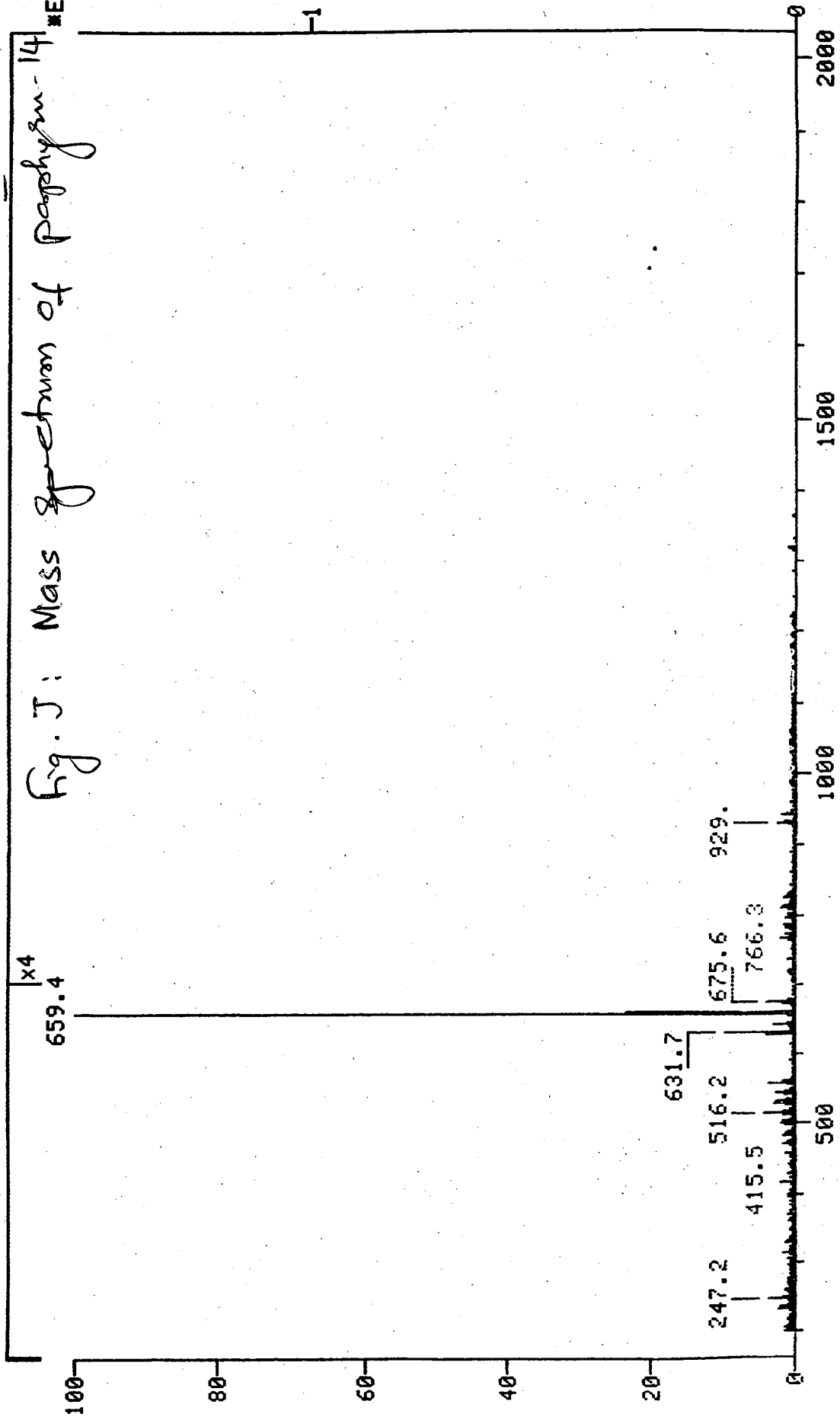
WinNmr - SEP35 001

14  
 SPEC: FP35  
 Samp: FP-35 (PANDEY)  
 Comm: POS. FAB\*\*m-NITROBENZYL ALCOHOL (1% TFA)  
 Mode: EI +VE +HMR BSCAN (EXP) UP LR  
 Oper: DUTTA  
 Base: 659.4  
 Norm: 659.4  
 Peak: 1000.00 mmu  
 Data: SY:FP35.DT

10-JUN-99 DERIVED SPECTRUM 9  
 Start : 08:39:19 10

Inlet :  
 Masses: 200 > 2000  
 # peaks: 1850

+ 'FP35' /1>10 - 'BLN61099 14





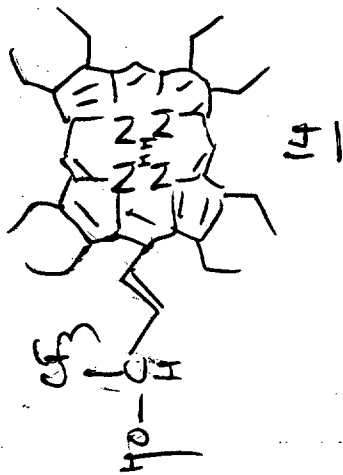
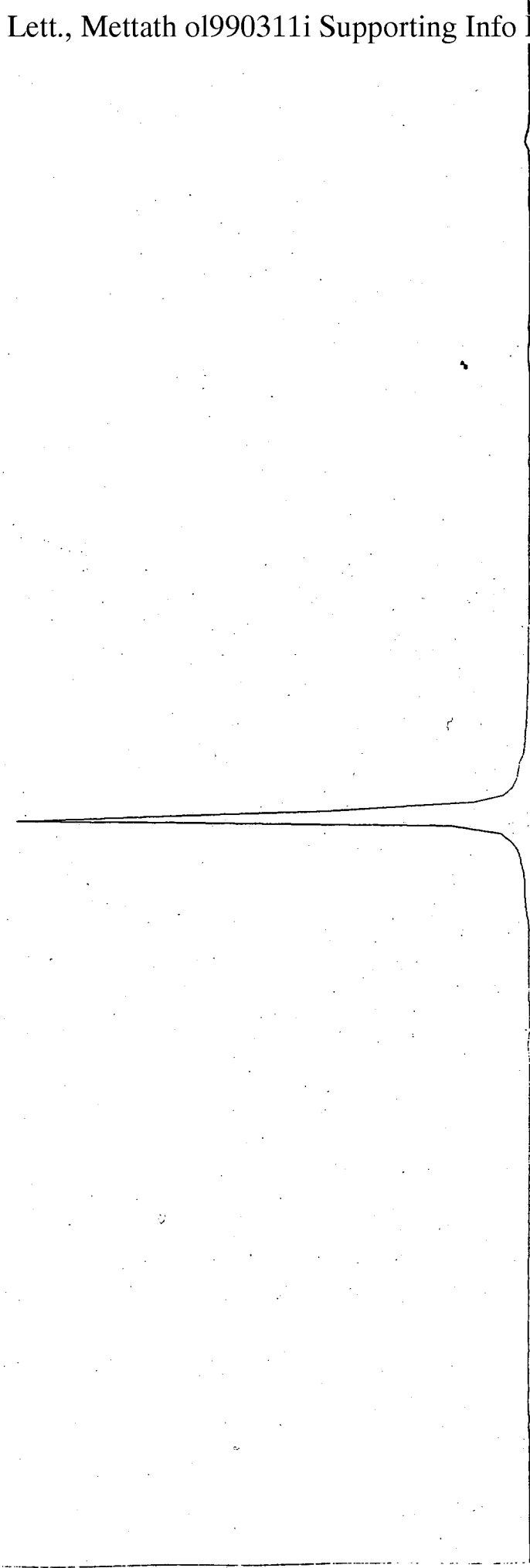


Figure k: <sup>19</sup>F NMR spectrum of porphyrin 14



77.6	-77.8	-78.0	-78.2	-78.4	-78.6	-78.8	-79.0	-79.2
CDC13, FP-35, F-19 NMR								
F1: 376.471			SW1: 26315			OF1: -24724.4		
EX			PW: 8.0 usec			NA: 48		
			PD: 2.0 sec			IJB: 2.0		
						PTSID: 8192		
						USER: -- DATE: 3/11/99		
						WinNuts - \$FIP35.001		

15a

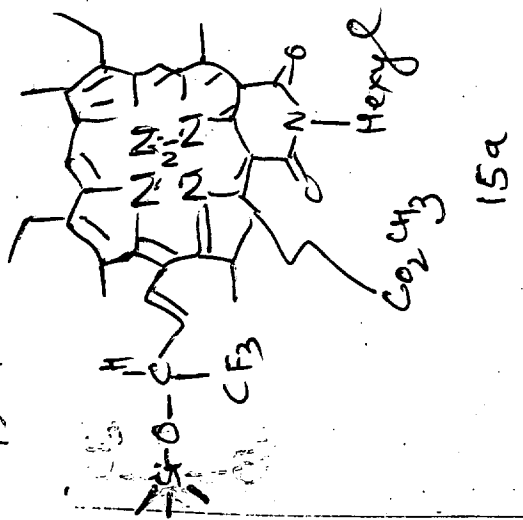
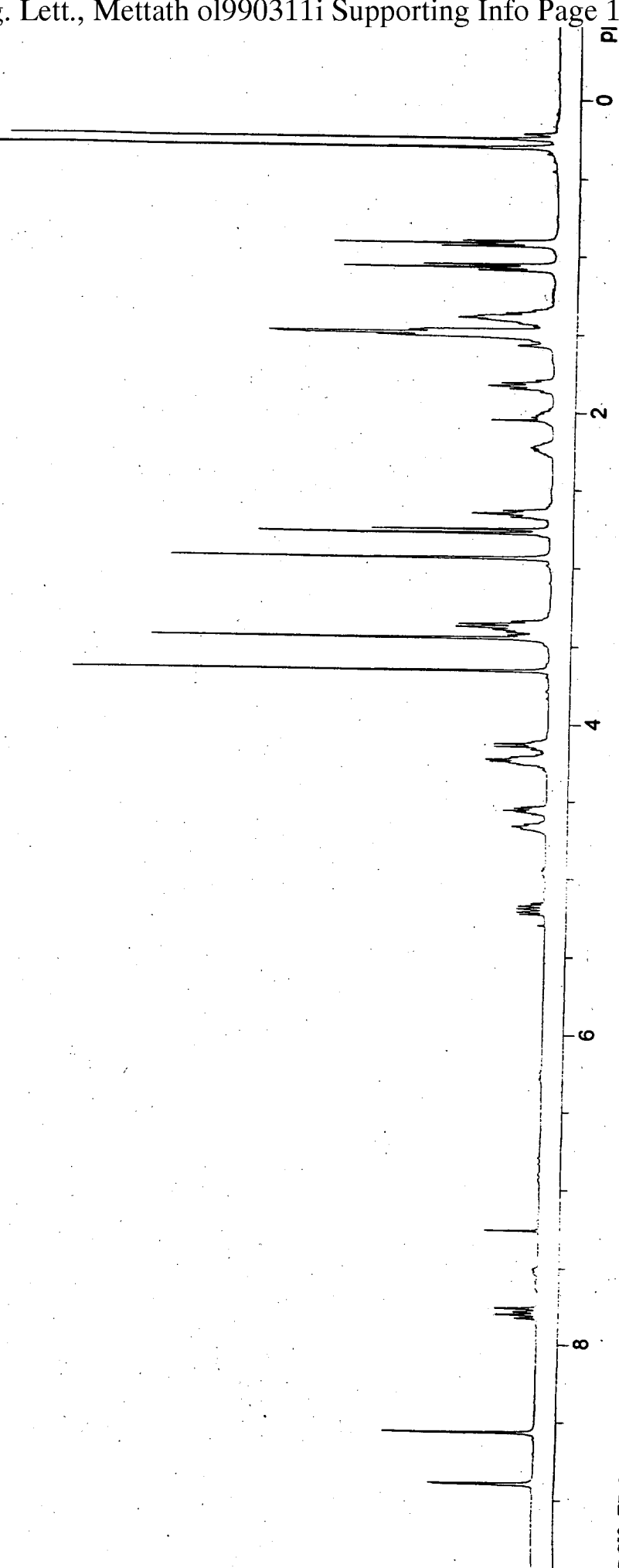


Figure L: <sup>1</sup>H NMR Spectrum of Chlorin 15a

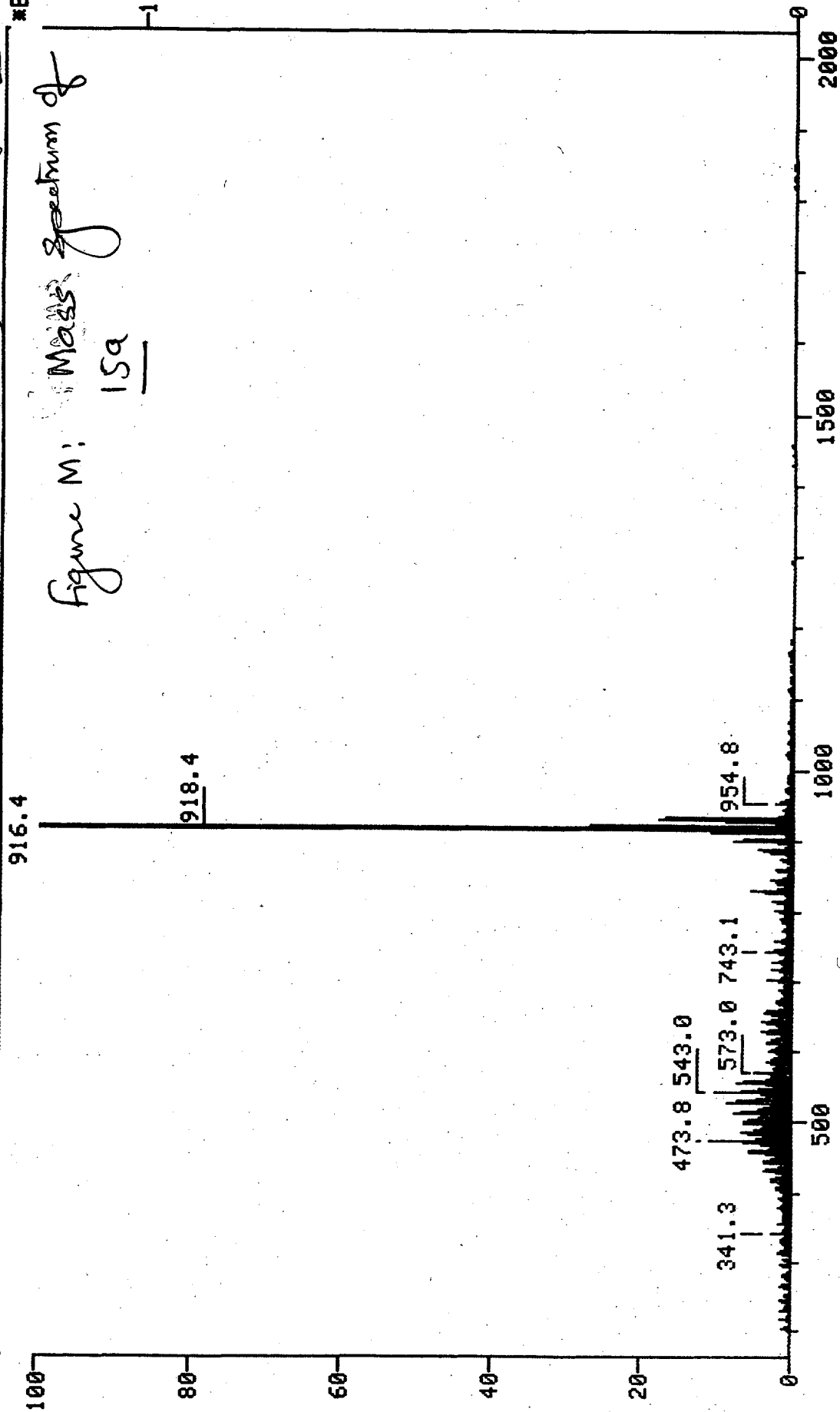
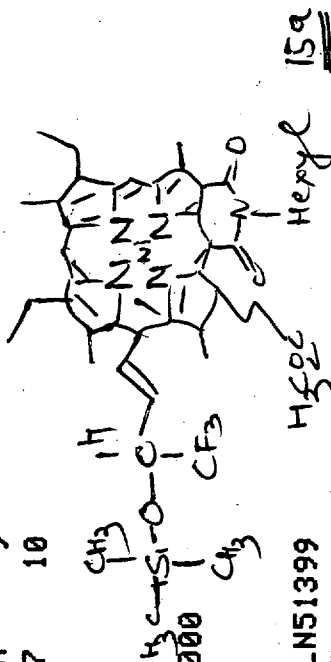


CDCI3, FP-23  
 F1: 400.138  
 EX:

SWI: 6024	OFI: 2207.3	USER: DATE: 06/03/99
PW: 7.0 usec	NA: 40	PTSId: 8192
PD: 40.0 sec	LB: 0.1	WinNuts - \$Fr23.001

15a

SPEC: FP23  
 Samp: FP-23 (PANDEY) 13-MAY-99 DERIVED SPECTRUM 9  
 Start : 13:13:37 10  
 Comm: POS. FAB\**m*-NITROBENZYL ALCOHOL  
 Mode: EI +VE +HMR BSCAN (EXP) UP LR  
 Oper: DUTTA  
 Base: 916.4 Inten : 115699968  
 Norm: 916.4 RIC : 2393896192  
 Peak: 1000.00 mmu  
 Data: SY:FP23.DT + 'FP23' /1>10 - 'BLN51399



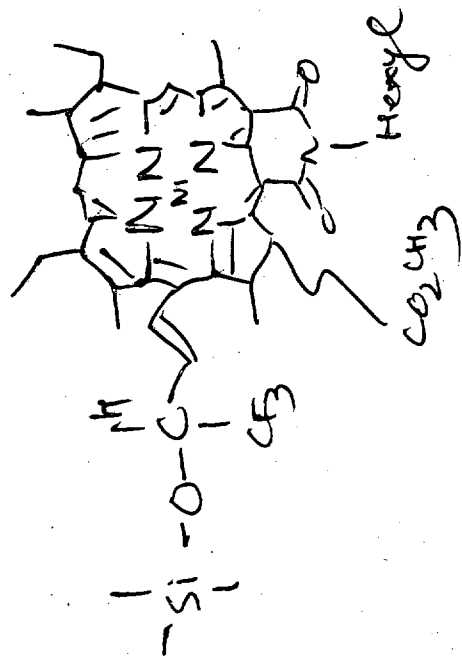
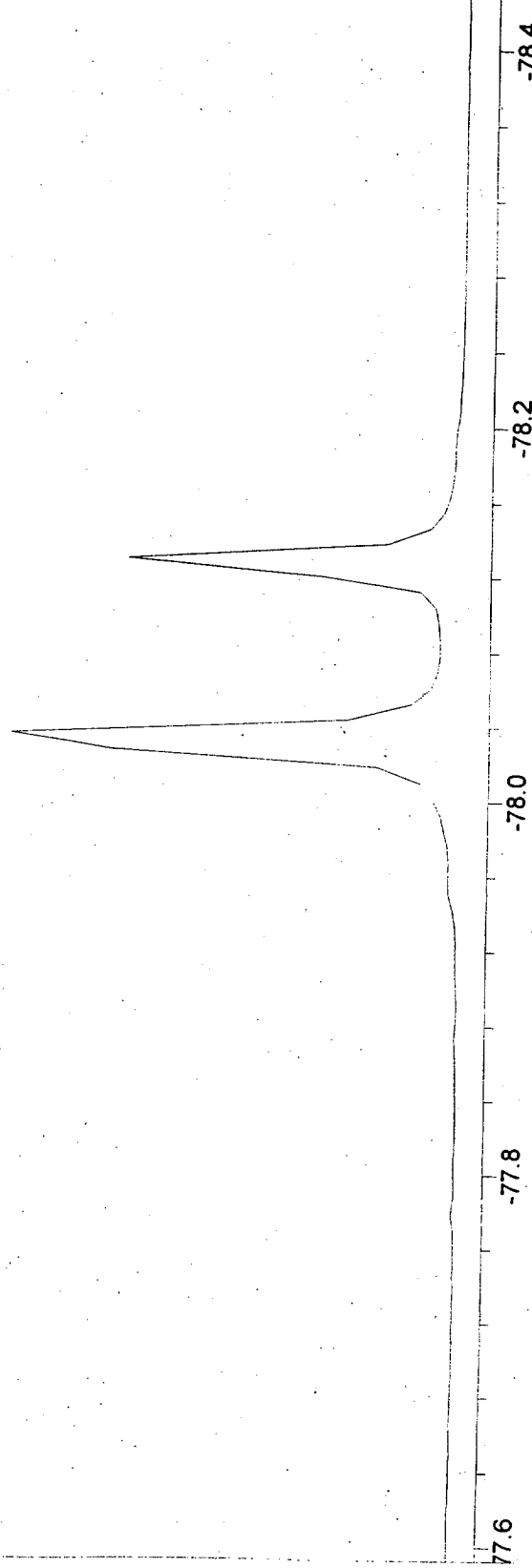


Figure N: <sup>19</sup>F NMR Spectrum of 15a



CDCl <sub>3</sub> , FP-23, F-19 NMR	SWH: 26315	OF1: -24724.4	PTSID: 8192	USER: DATE: 06/03/99
F1: 376.471	PW: 8.0 usec	NA: 40	LB: 2.0	WinNuts - \$FFm23 001
EX:				

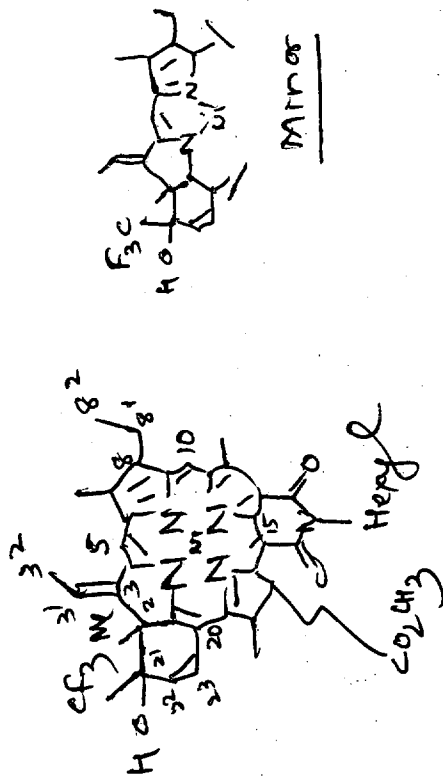
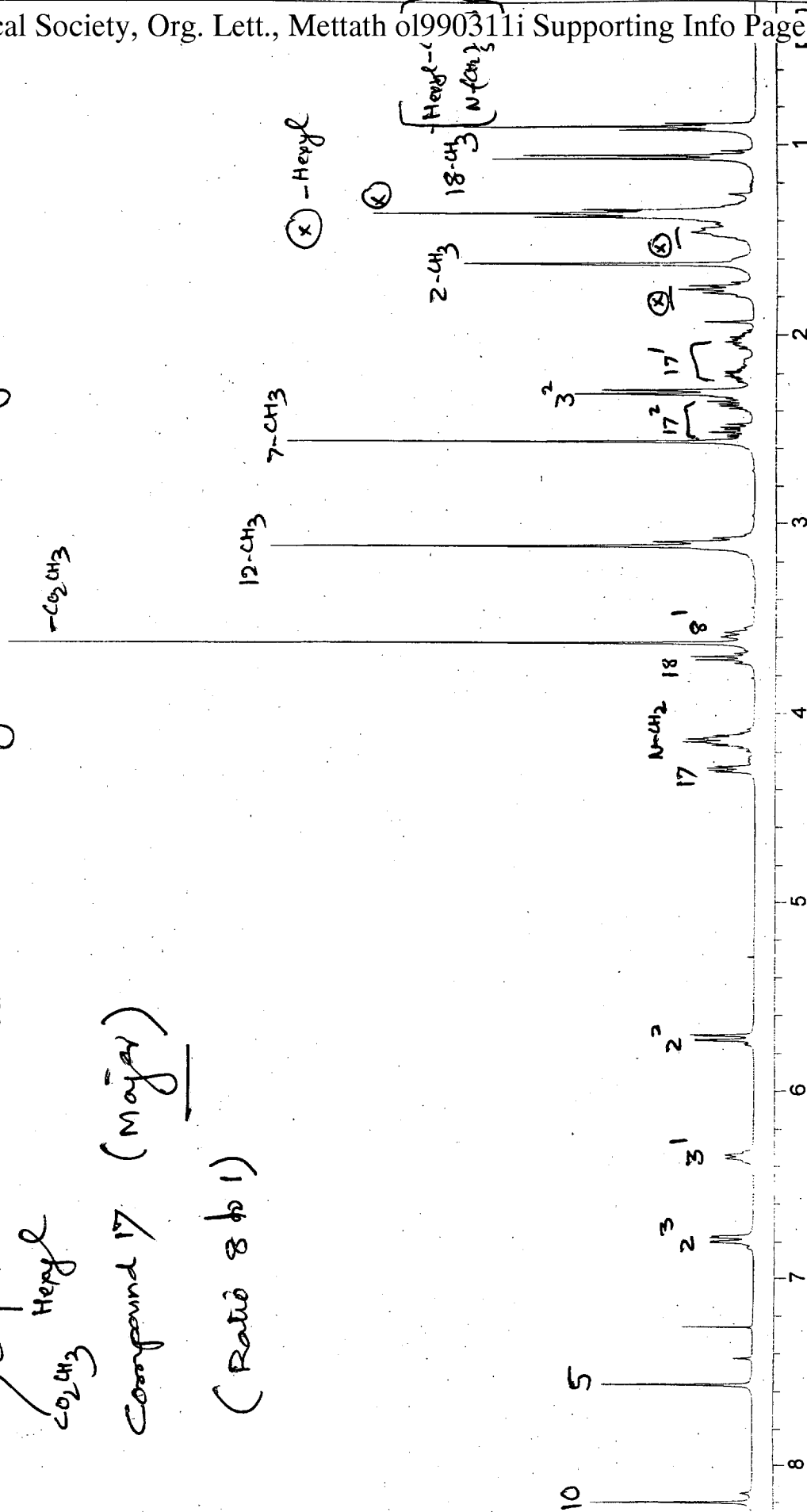
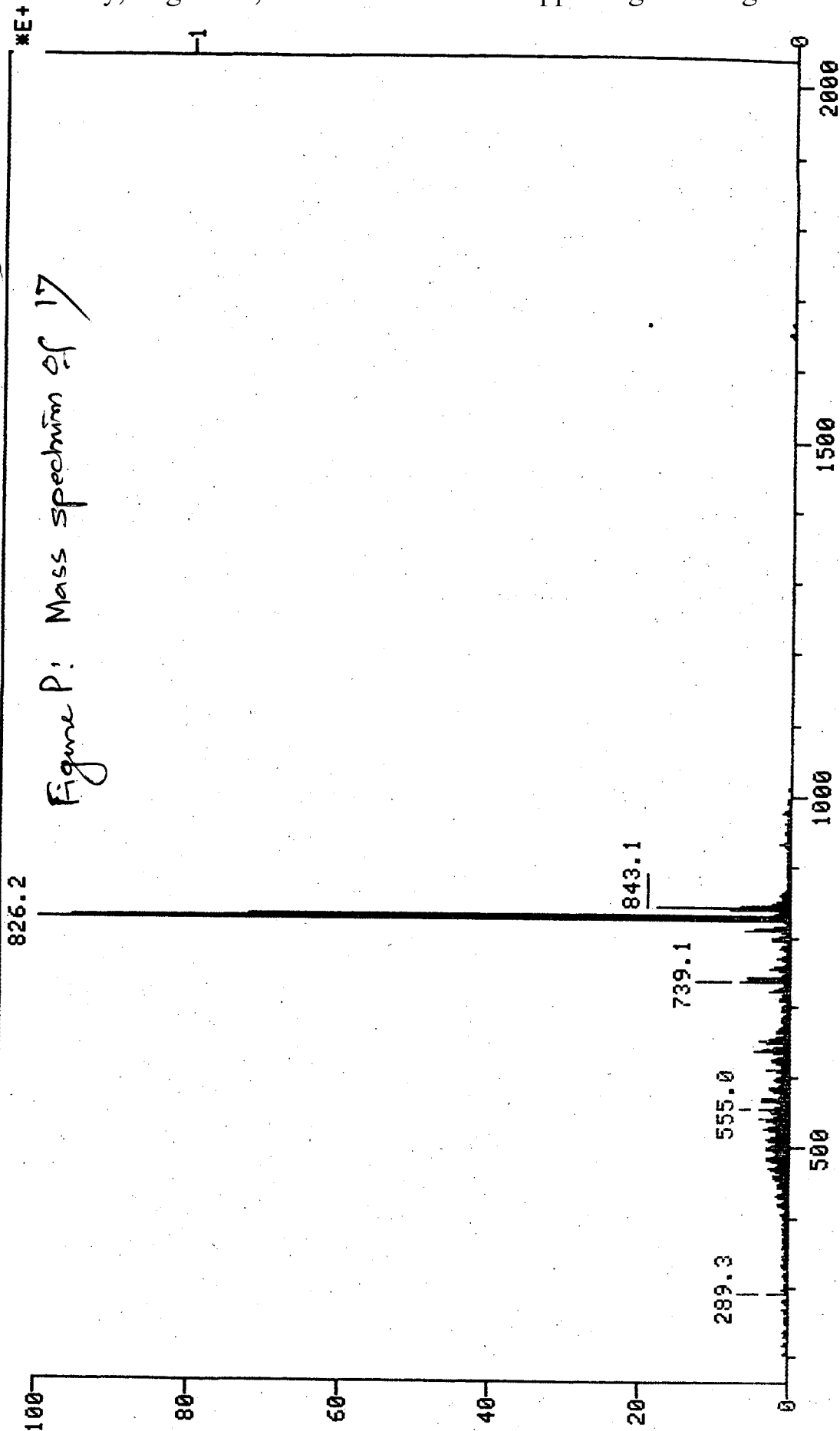
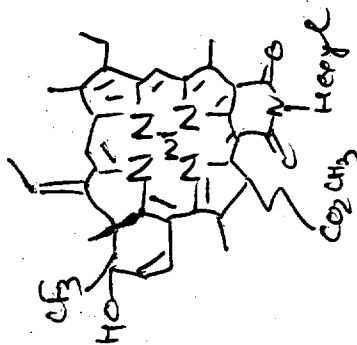


Figure 0: <sup>1</sup>H NMR Spectrum of 17

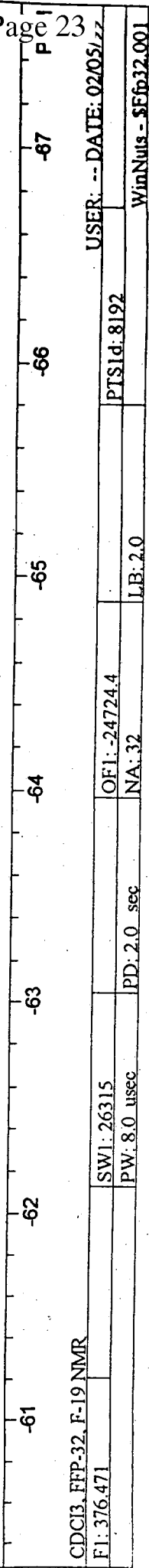
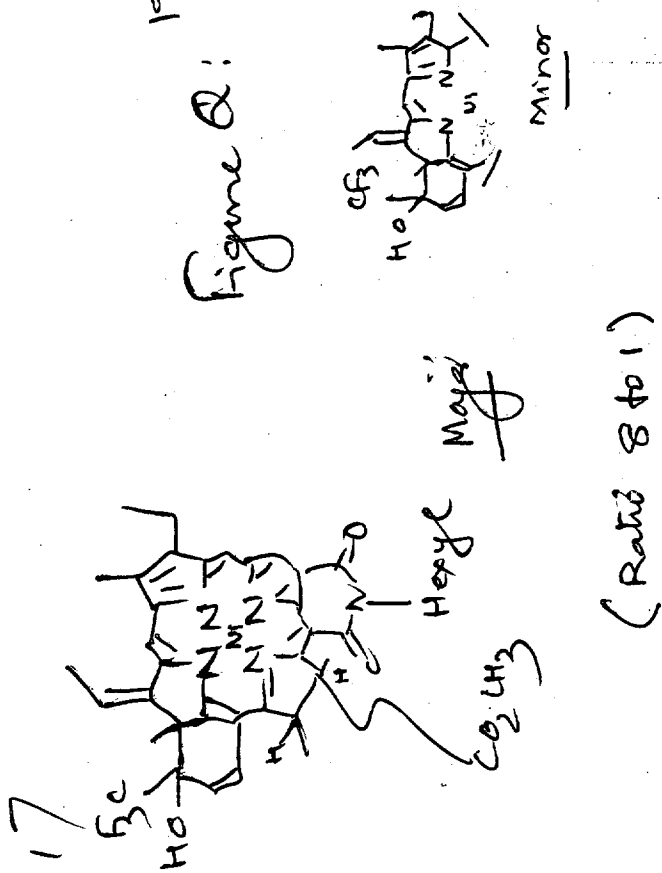


Compound 17 (Major)  
(Ratio 8 to 1)

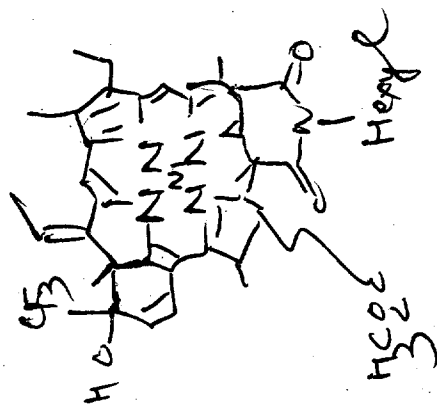
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 Samp: FP-32 (PANDEY)  
 Comm: POS. FAB\**m*-NITROBENZYL ALCOHOL  
 Mode: EI +VE +HMR BSCAN (EXP) UP LR  
 Oper: DUTTA  
 Base: 826.2  
 Norm: 826.2  
 Peak: 1000.00 mmu  
 Data: SY:FP32.DT  
 13-MAY-99 DERIVED SPECTRUM 9  
 Start : 10:17:30 10  
 Inlet :  
 Masses: 200 > 2000  
 # peaks: 1980  
 + 'FP32' /1>10 - 'BLN51399'



17  
 Figure 2:  $^{19}\text{F}$  NMR Spectrum of 17 (Z/E Isomers)

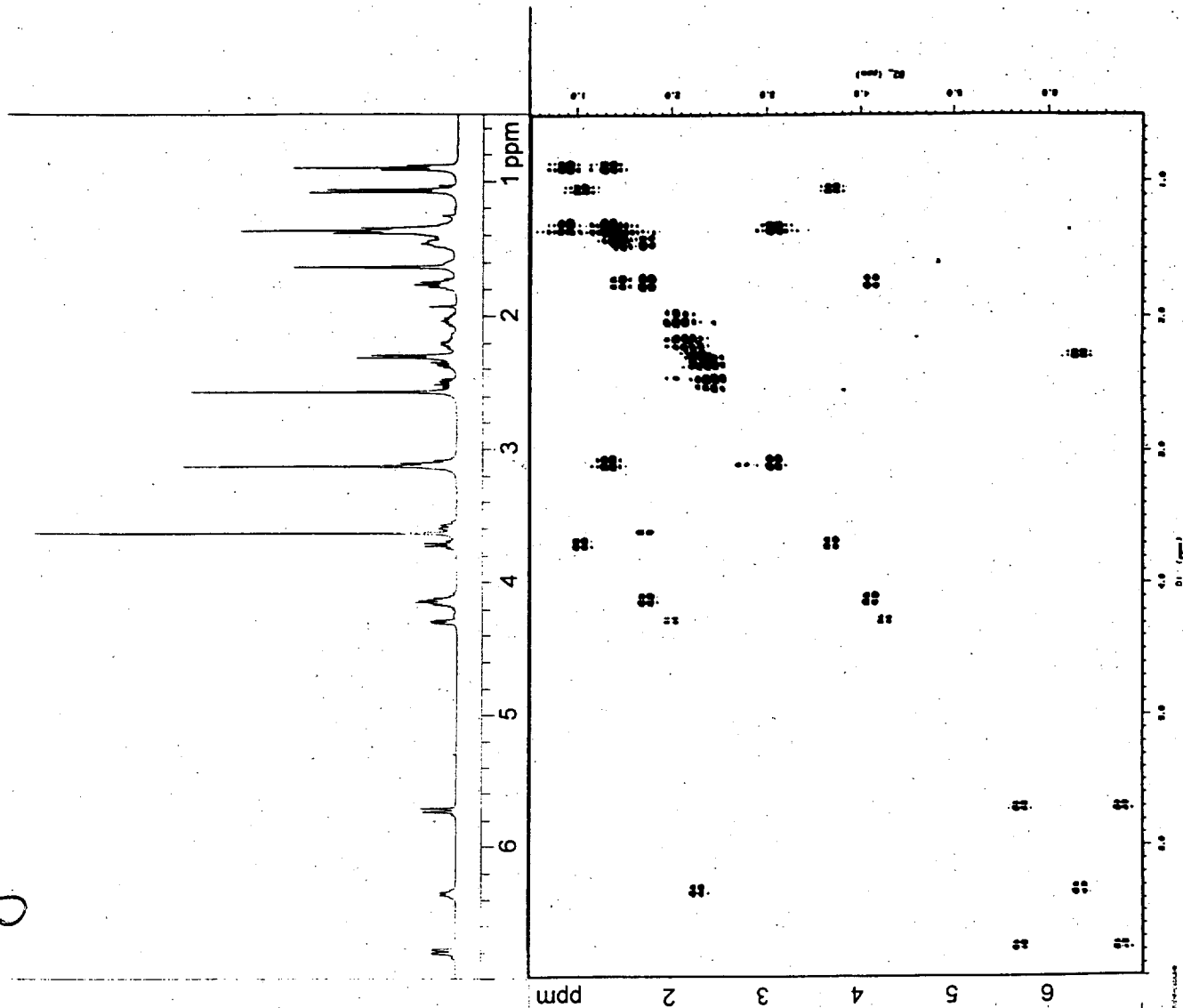


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F1: 376.471	PW: 8.0 usec	NA: 32	LB: 2.0	WinNuts - \$Efp12.001
	PD: 2.0 sec			

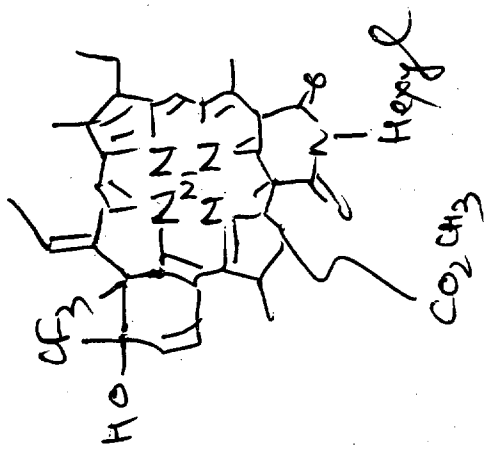


ISOBACTERIOCHLORIN 17

Fig. R1: COST DATA

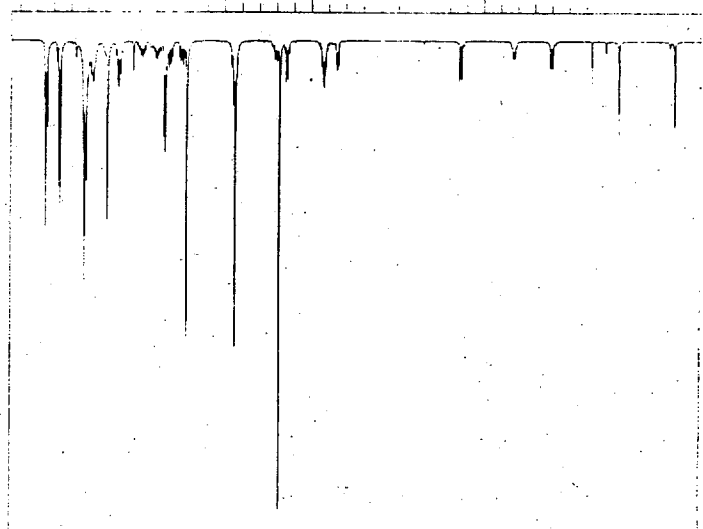
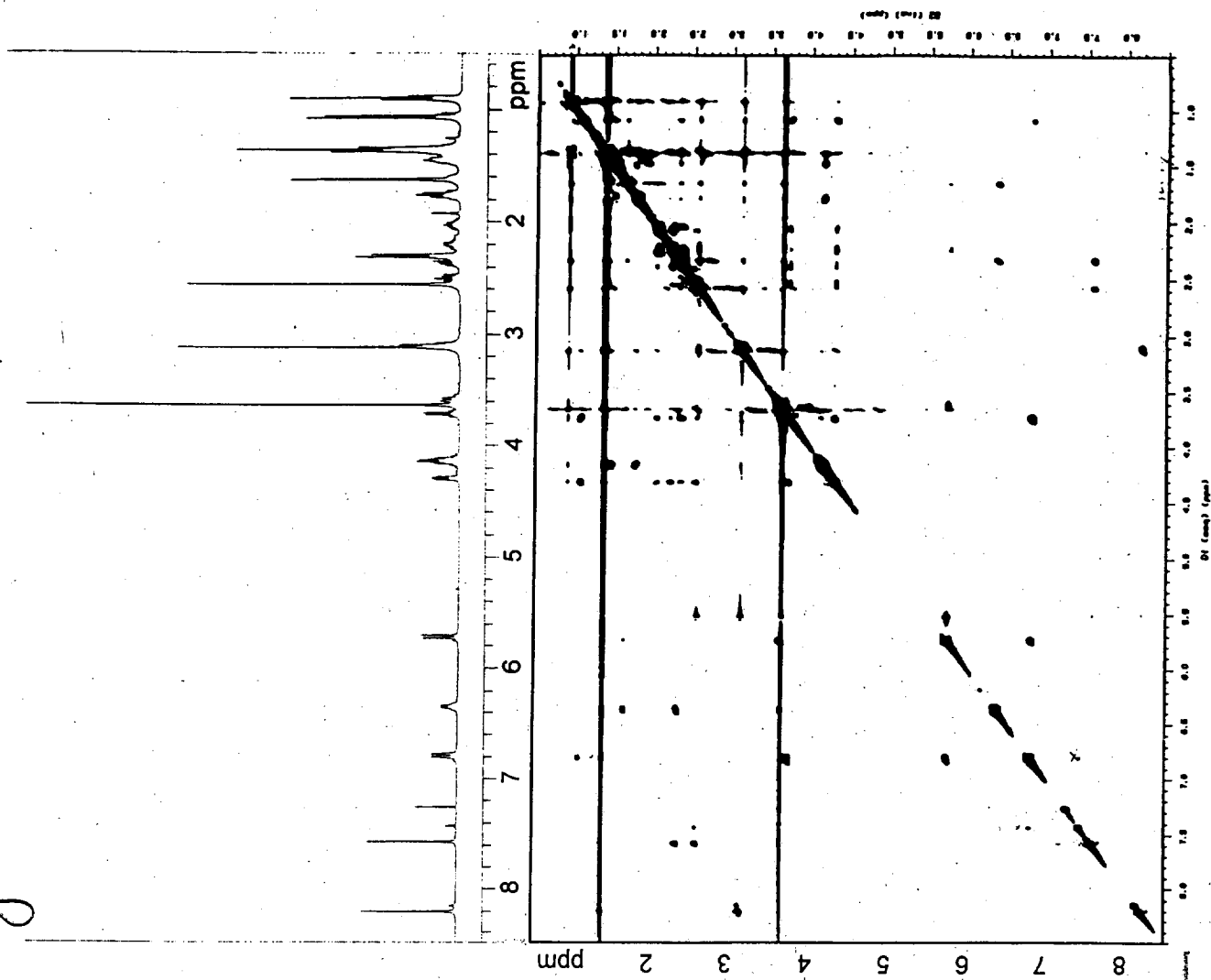






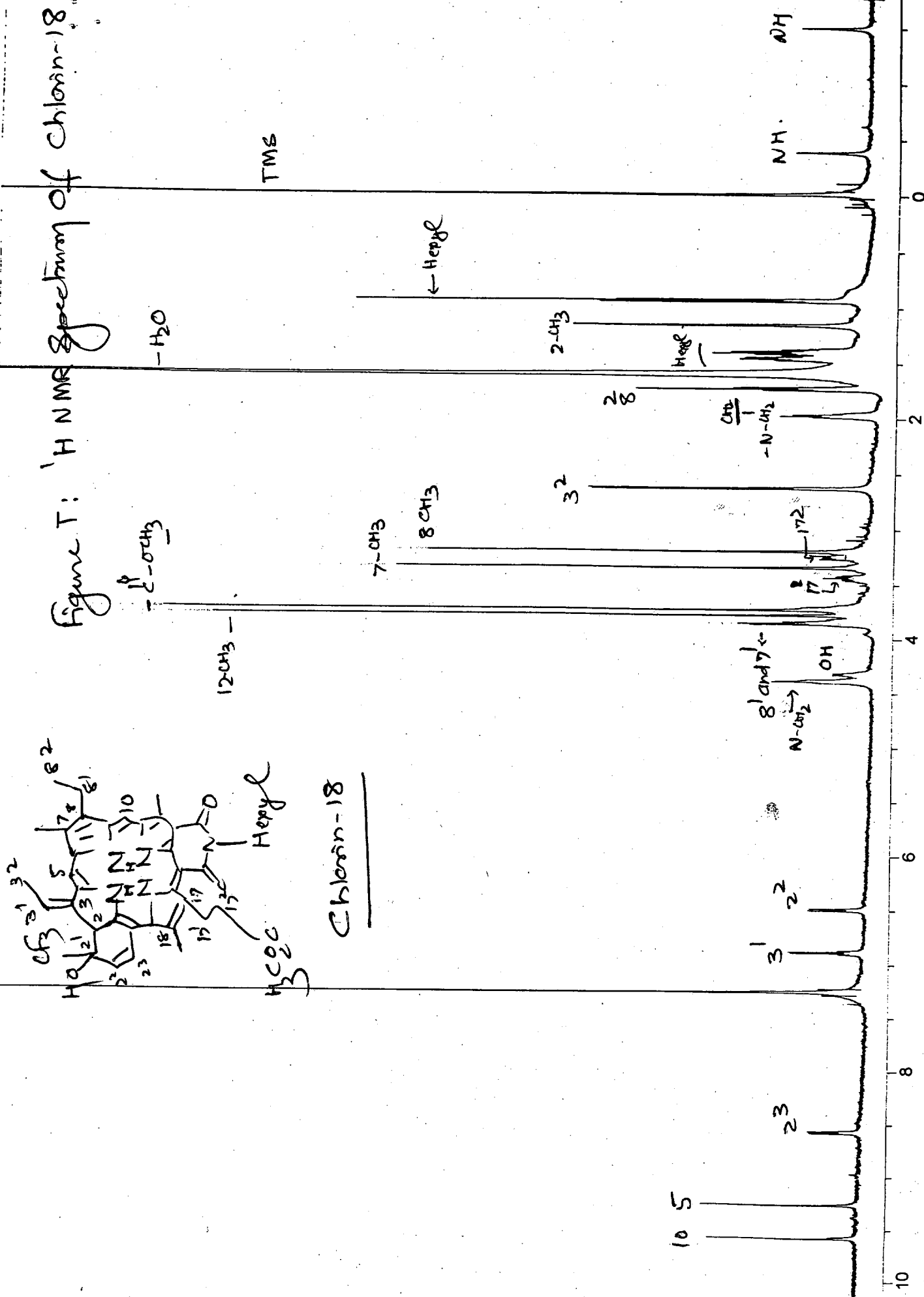
ISOBACTERIOCHLORIN 17

Figure 5: ROESY DATA



18, 600 MHz

Figure T: <sup>1</sup>H NMR Spectrum of Chlorin-18



FI: 600.139	SW1: 8065	OF1: 2267.3	PTSID: 32768	USER: DATE: 00/00
EX: lul	PW: 0.0 usec	PD: 0.0 sec	LB: 0.1	WinNmr - 60231 600.001

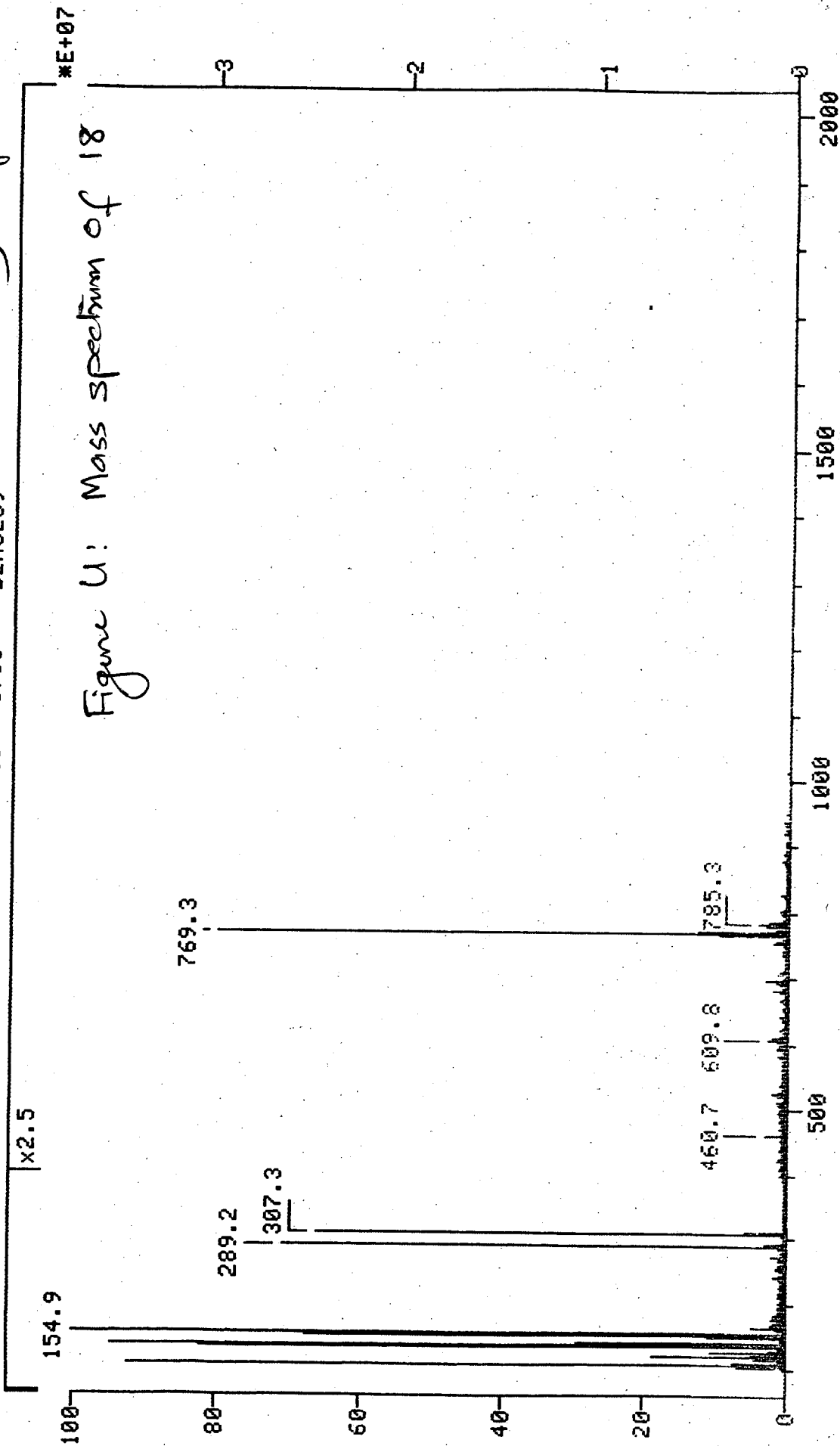
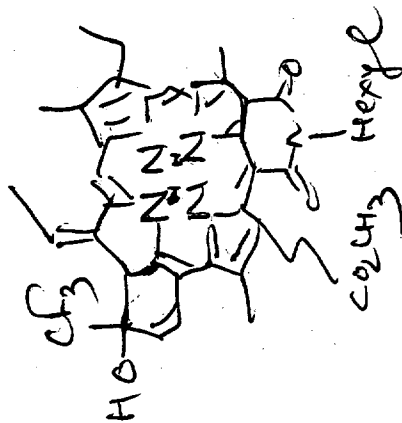
18

SPEC: FP231  
 Samp: FP-231 (PANDEY)  
 Comm: POS. FAB\*m-NITROBENZYL ALCOHOL  
 Mode: EI +VE +HMR BSCAN (EXP) UP LR  
 Oper: DUTTA  
 Base: 154.9  
 Norm: 154.9  
 Peak: 1000.00 mmu  
 Data: SY:FP231.DT

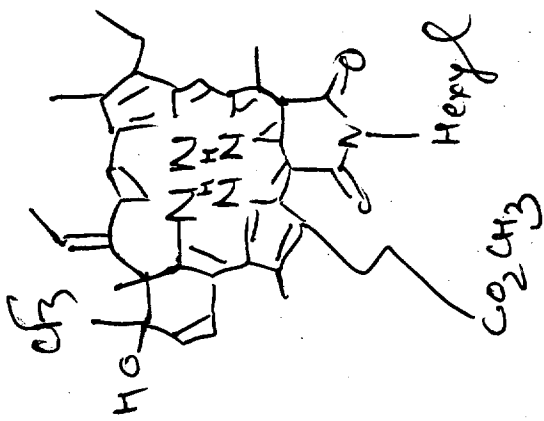
20-AUG-98 DERIVED SPECTRUM 9  
 Start : 11:02:48 10

Inlet :  
 Masses: 100 > 2000  
 # peaks: 1394

+ 'FP231' /1>10 - 'BLN8209

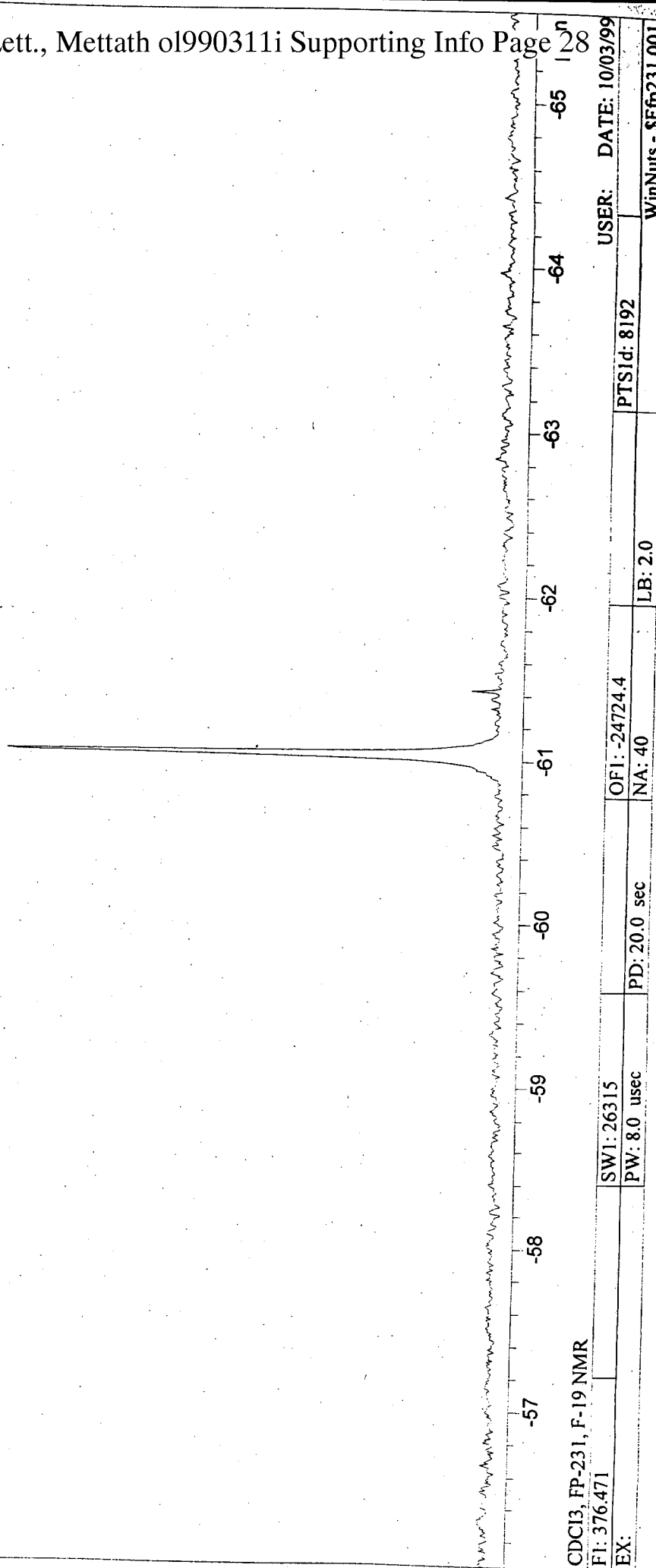


18



Chlorin 18

Figure V: <sup>19</sup>F NMR Spectrum of Chlorin 18



CDC13, FP-231, F-19 NMR  
 FI: 376.471  
 EX:

SW1: 26315	OF1: -24724.4	PTSID: 8192	USER: DATE: 10/03/99
PW: 8.0 usec	NA: 40	LB: 2.0	WinNuts - \$Ffb231.001
PD: 20.0 sec			



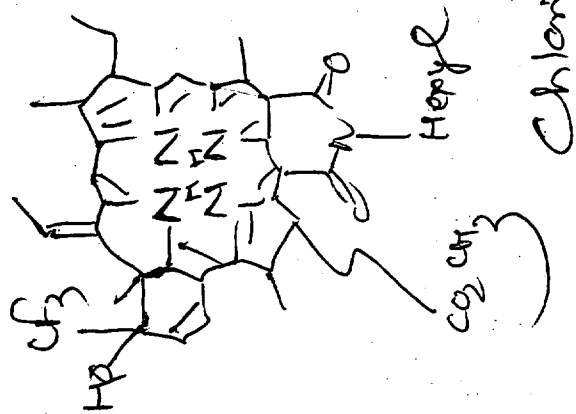


Figure X: ROESY DATA

